

# Geochemical controls on mobilization of metals from tailings and implications for cover amendments

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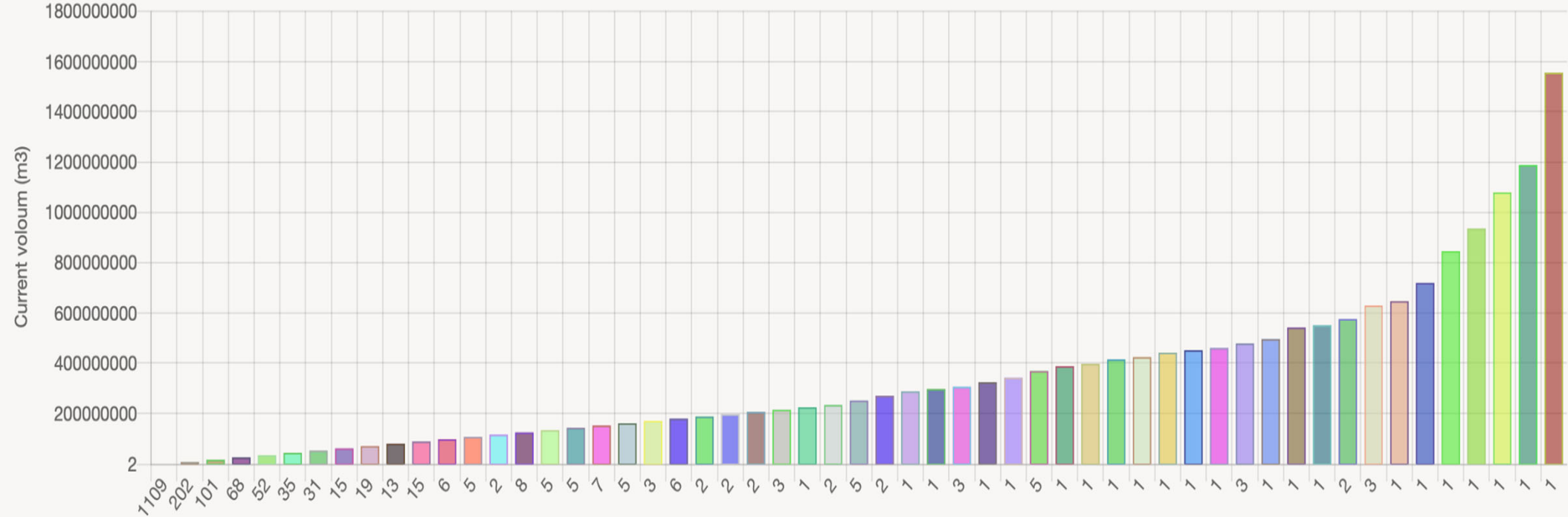
# World-wide tailings facilities location reported by Industry



<https://tailing.grida.no/map/maps/>

# World-wide tailings facilities volume reported by Industry

Number of Tailings storage facilities based on current storage volume interval (9 Million)



<https://tailing.grida.no/map/overview/>



# Motivation

## Potential Environmental impacts from Tailings, Waste Rock and Mine Drainage

Fresh rock surfaces are exposed to atmospheric conditions

Oxidation & mineral dissolution cause metals-laden drainage

Drainage often requires treatment

Dust produced from erosion mobilize by wind and human activities



# Metals Mobilization Reported from Typical Amendments

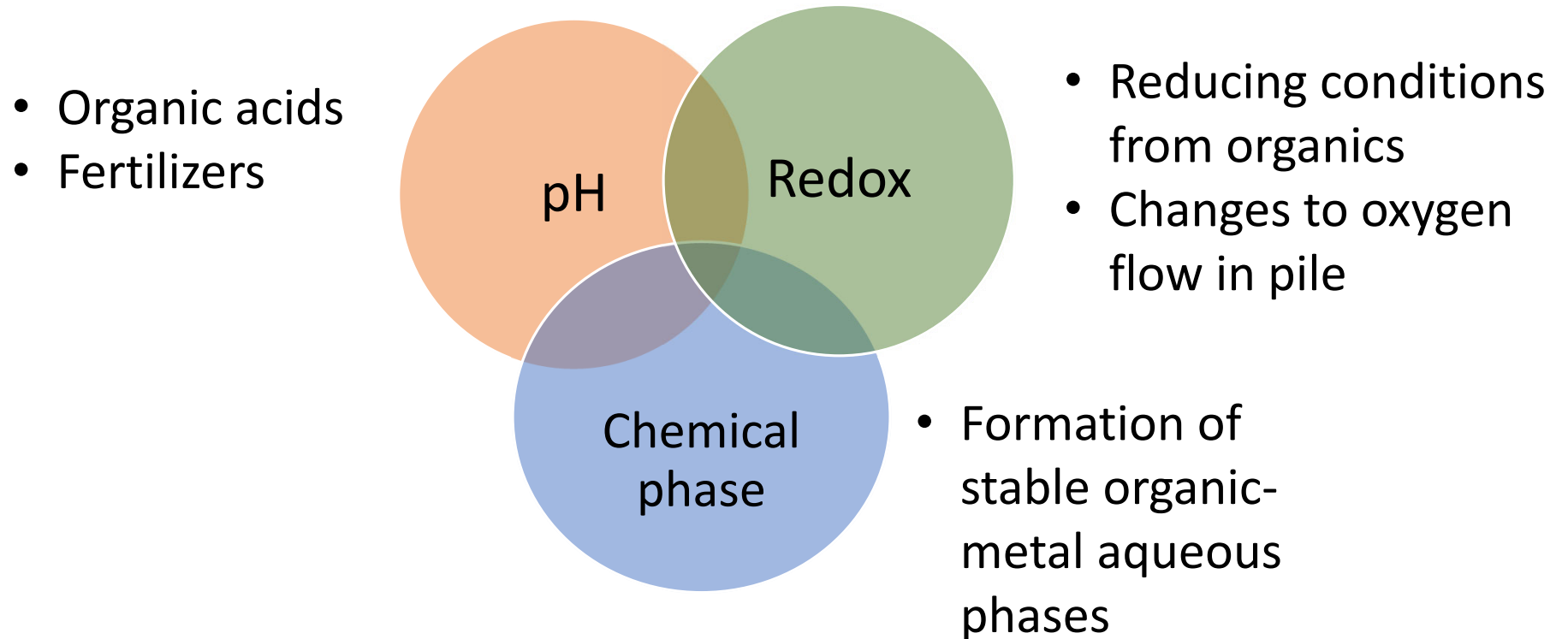
Amendment	Metals Mobilized				Reference
	Cu	Cd	Pb	Zn	
Sewage sludge	✓				Andrés and Francisco 2008
Sewage sludge				✓	Andrés and Francisco 2008
Sewage sludge		✓	✓		Harwood et al. 1986
Leaf cover			✓		Harwood et al. 1986
Composted yard waste		✓	✓	✓	Schwab et al. 2006
Composted cattle manure		✓	✓	✓	Schwab et al. 2006
Aged cattle manure		✓	✓	✓	Schwab et al. 2006

Andres, NF and Francisco, MS, 2008, Effects of sewage sludge application on heavy metal leaching from mine tailings impoundments, *Bioresource Technology*, Vol. 99, Issue 16.

Harwood, JJ, Koirtyohann, SR, and Schmitt, CJ, 1986, Effects of Cover materials on leaching of constituents from dolomitic lead mine tailings, *Water, Air, and Soil Pollution*, 34, pp 31-43.

Schwab, P, Zhu, D, Banks, MK, 2007, Heavy metal leaching from mine tailings as affected by organic amendments, *Bioresource Technology*, Vol. 98, pp. 2935-2941.

# Goal: Characterize Metal Mobilization from Potential Geochemical Changes Induced by Cover Amendments



# Methods to Characterize Metal Mobilization Potential of Waste

- Elemental Composition  
Microwave assisted acid-digestion, followed by ICP-AES
- Sequential Extractions (Tessier et al. 1979)

Fraction 1	Fraction 2	Fraction 3
<ul style="list-style-type: none"><li>• Exchangeable</li><li>• 1 M MgCl<sub>2</sub> pH 7</li></ul>	<ul style="list-style-type: none"><li>• Carbonate Bound</li><li>• 1M sodium acetate pH 5</li></ul>	<ul style="list-style-type: none"><li>• Fe and Mn oxide bound (reducible)</li><li>• 0.04M hydroxylamine hydrochloride in 25% acetic acid</li></ul>

\*Modified Tessier et al. 1979 method, with F4 (bound to organic carbon) omitted

# Amendment Extraction Methods

- Leach tests with amendment solutions
  - 1 g rock + 20 mL extraction solution
- Extraction of Waste Rock + Amendments
  - 1 g rock + 0.5 g amendments
  - 1 M MgCl<sub>2</sub> at pH 7
- Analysis
  - Filter 0.45 microns
  - ICP-OES

## Amendments used in Leach Tests

Amendment(s)	Source	Solution Description	Initial pH of Solution
None	--	Deionized Water	--
Topsoil	EarthGro Topsoil	Extraction*	7.1
Spent Brewery Grain	Green Mountain Beer Company, Lakewood, CO	Extraction*	3.8
Biochar	Biochar Solutions, Lafayette, CO	Extraction*	9.1
Compost	3XM Grinding & Compost, Olathe, CO	Extraction*	8.4
Compost/Biochar	Biochar Solutions and 3XM Grinding & Compost	Extraction*	8.5
Soil Media**	Proganics™ Biotic Soil Media (Profile 2020)	Extraction*	6.8
Phosphate	Calcium phosphate monobasic monohydrate	4 mmol P/L	5.7

\*Extraction Solution prepared by drying solids for 48 hours at 40 degrees C, followed by extraction with deionized water for 48 hours at room temperature and constant mixing.

\*\* Soil Media is Proganics™ Biotic Soil Media, produced by Profile Products



# Waste Material Composition

## Average Elemental Composition (% by mass)

Al	Ca	Cu	Fe	K	Mg	Mn	P	Pb	S	Zn
1.3	2.5	0.75	4.4	0.45	0.33	1.3	0.12	3.5	1.0	0.80

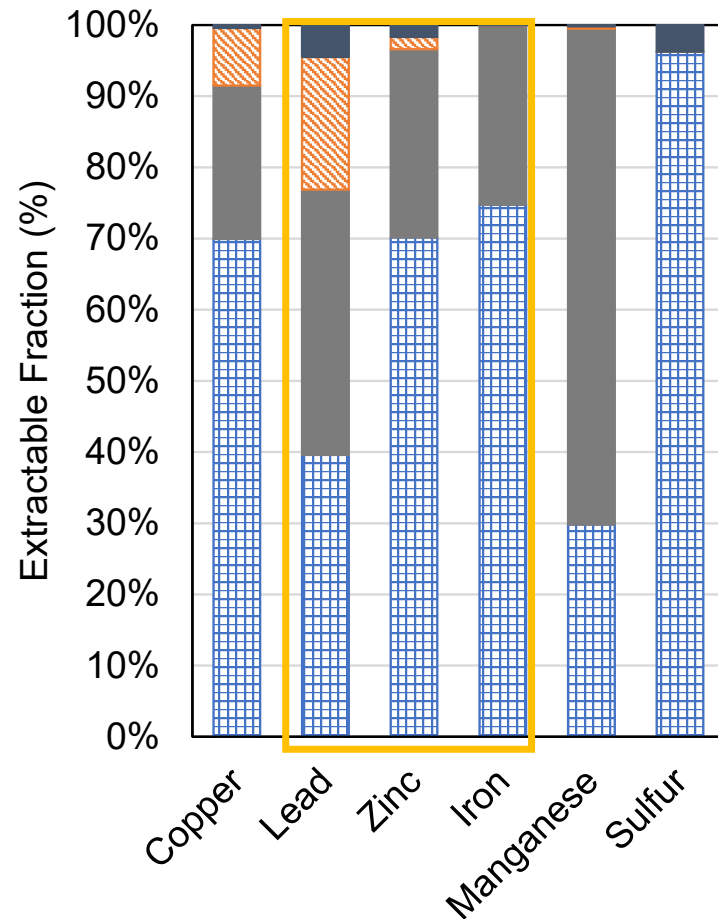
**35,000 mg/kg Pb:** in waste material

**1,000 mg/kg Pb:** typical cleanup level for industrial site (EPA 2015)

**1.8 mg/L Pb:** concentration if 1% of lead mobilized (20:1 water to waste material)

**0.01 mg/L Pb:** Chronic aquatic life standard (hardness of 400 mg/L, CCR 1002-34)

# Metal Fraction Association

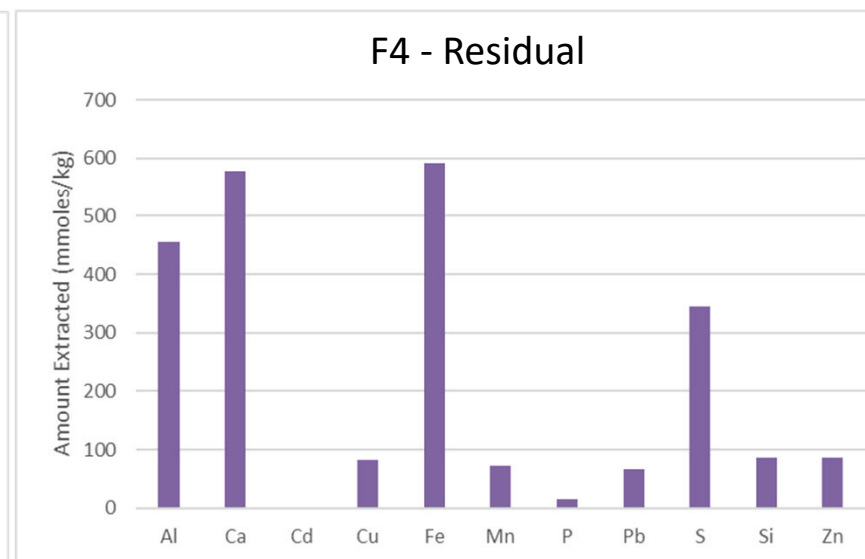
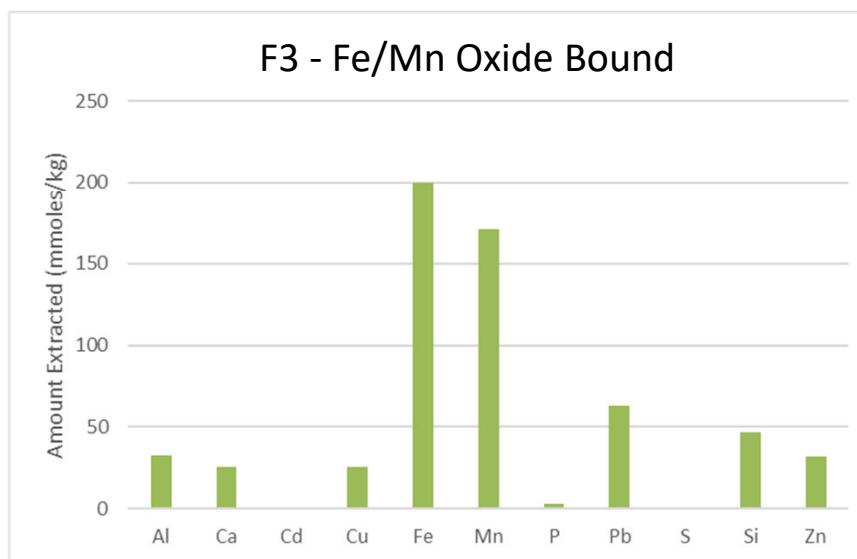
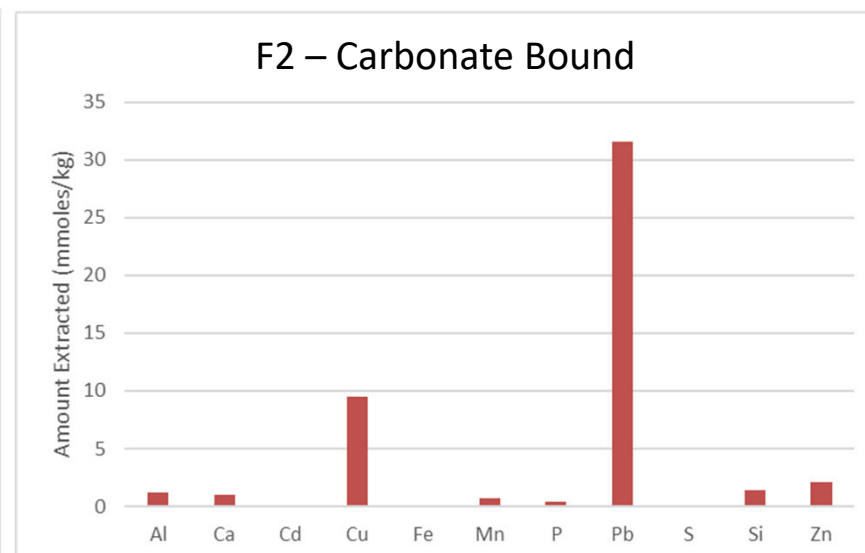
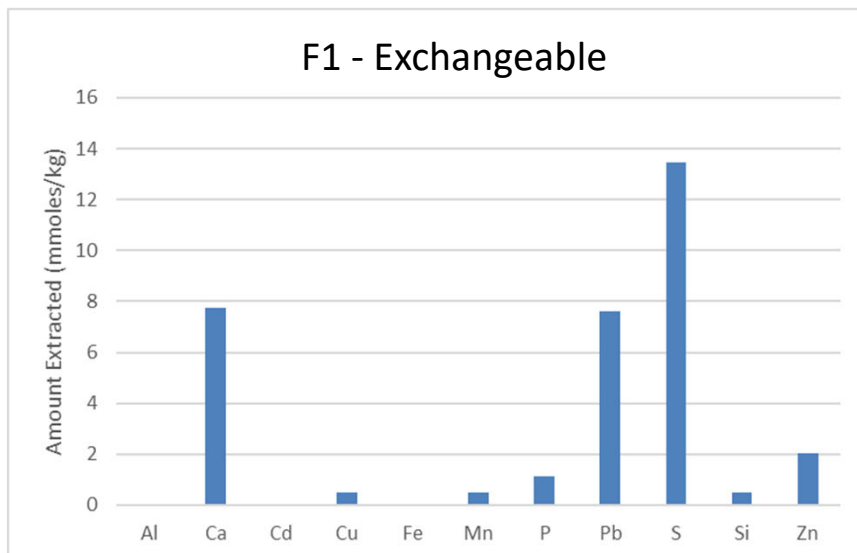


- Potential weathering impact
- Large fraction associated with Fe & Mn oxides:
  - 22% of Copper
  - 37% of Lead
  - 26% of Zinc

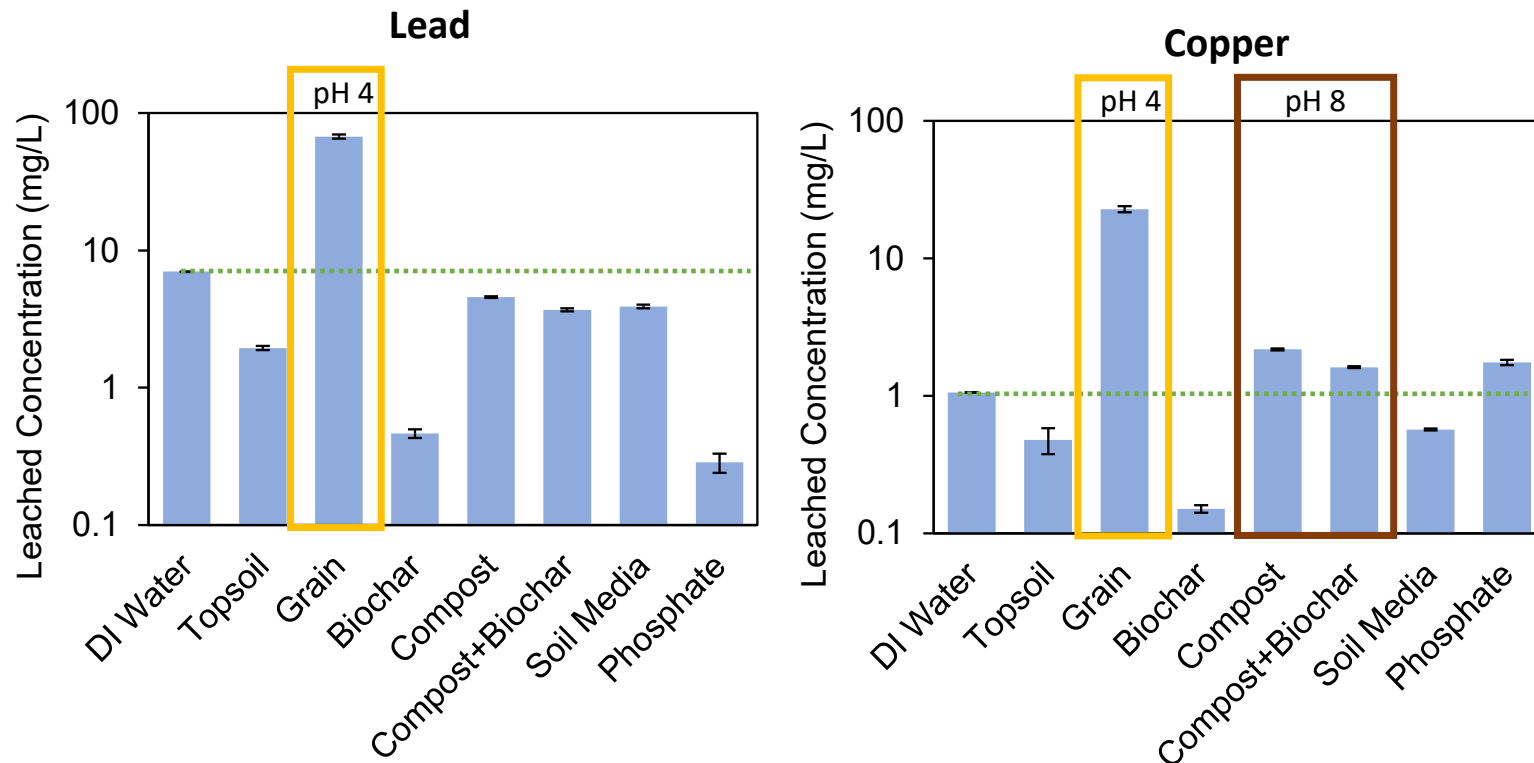
- F1 - Exchangeable
- ▨ F2 - Carbonates
- F3 - Mn/Fe Oxides
- ▤ Residual

**Cover amendments that produce strongly reducing conditions will mobilize metals**

# Metal Mass Association



# Metals Mobilization by Amendment Leachate



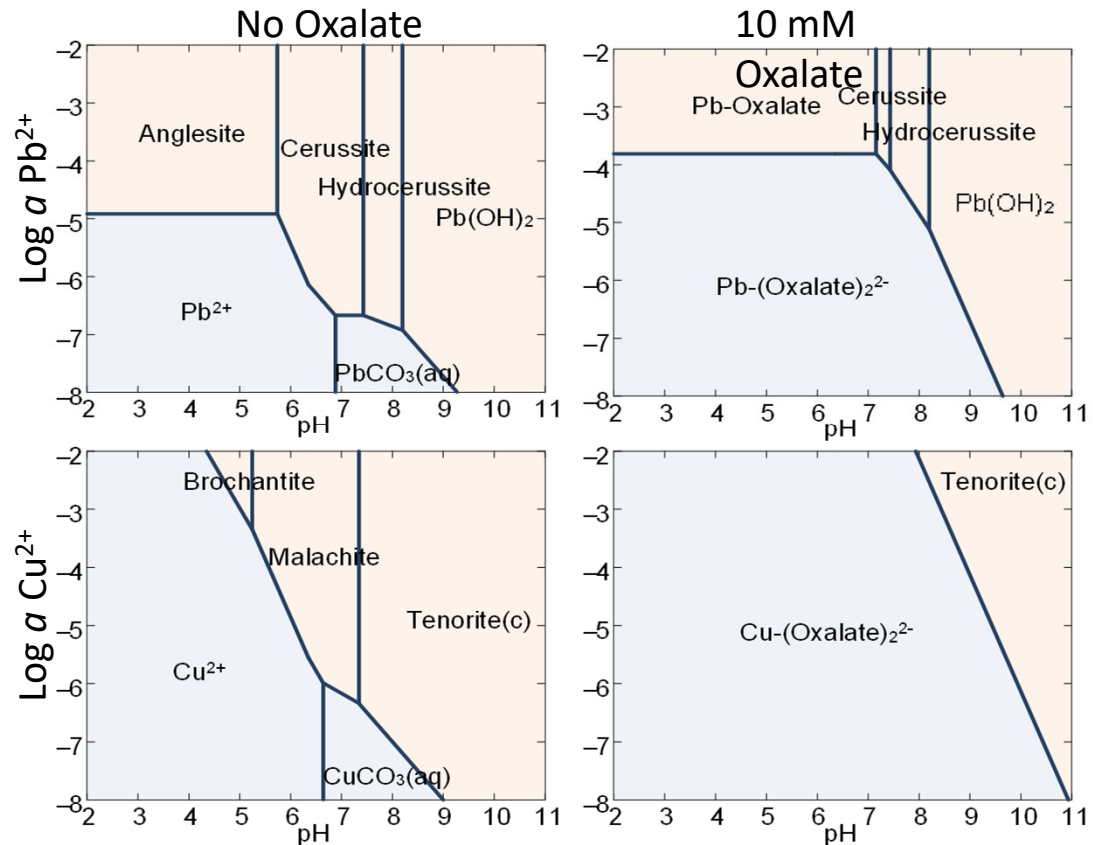
Low pH of brewery grain caused dissolution of cerussite/lead phosphates and desorption of metals from oxides

# Controls on Metals Mobilization: Aqueous Metal-Organic Complexes

Formation of  
stable aqueous  
metal-organic  
complexes  
possible

Copper complexes  
are more stable  
than for Pb

Contribute to Cu  
mobilized by  
compost



Diagrams created with Visual MINTEQ database  
formatted for Geochemist Workbench

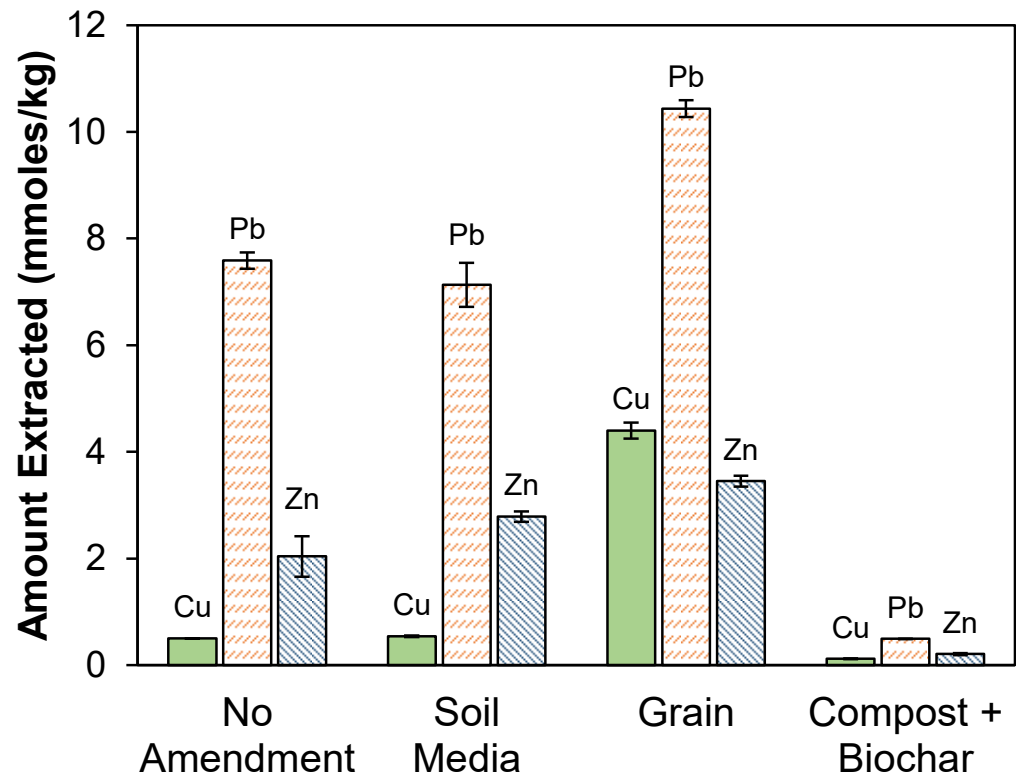


# Controls on Metals Mobilization: Waste/Amendment Extracts

Biochar + compost decreased exchangeable fraction of Pb, Cu, and Zn by 87% (avg.)

Soil media did not reduce metals mobilization relative to DI control

Brewing grain increased Cu, Zn and Pb leaching compared to other treatments



# Acknowledgements

- James Ranville, Colorado School of Mines
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- Green Mountain Beer Company

# Treatment Implications for Tailings and Waste Rock

Identifying metal phases  
critical

Sequential extraction  
effective for waste  
characterization

Extraction with  
waste/amendment mixture  
critical

Knowledge of metal phases  
allows for more effective  
field trials

Organic amendments with  
low pH and high soluble  
carbon present the  
greatest risk





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