

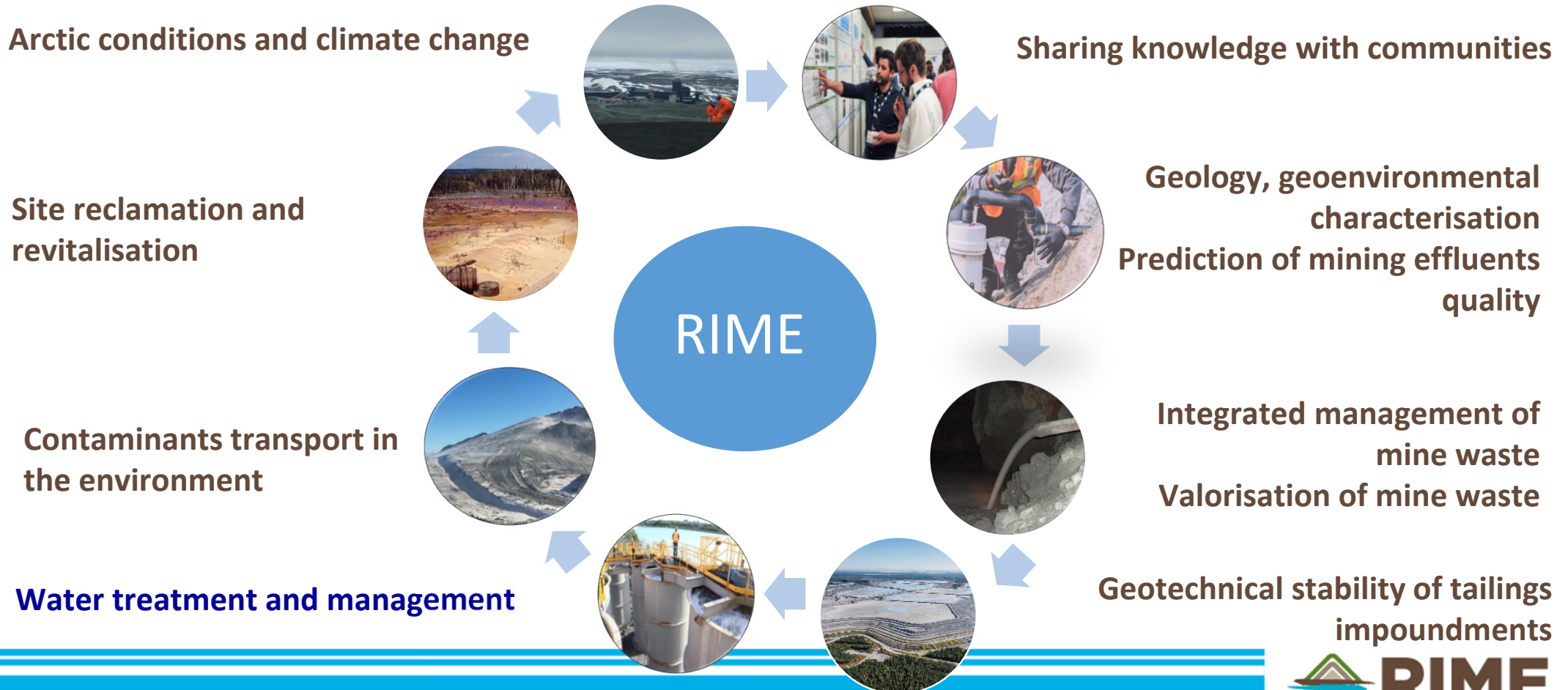


# Performance of natural and residual materials for mine water treatment and mine sites rehabilitation

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# Research topics



<https://www.uqat.ca/uqat/departements/irme/>

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# Industrial partners mine sites



<https://www.uqat.ca/uqat/departements/irme/>

# Introduction: Mines and water contamination

Contaminants	Mine drainage (runoff water)		N-based compounds (mine effluents)	
	AMD (acid mine drainage)	NMD (neutral mine drainage)	CN <sup>-</sup> , SCN <sup>-</sup>	Ammonia (NH <sub>3</sub> -N)
<i>Sources</i>	Metal sulfides + O <sub>2</sub> + water		Au, Ag extraction + blasting agents (ANFO)	
<i>Characteristics</i>	<b>pH &lt; 3; high [] metals (Fe &gt;1g/L), sulfates</b>	<b>Metal [] &gt; criteria</b>	Ore dependent, <b>but [] &gt; criteria</b>	
<i>Why prevent or treat?</i>	Regulation, environmental and social impacts			
<i>Challenges</i>	<b>Several contaminants</b>	<b>High contaminant mobility</b>	<b>Complexity (toxicity, costs, flowrates)</b>	
<i>Treatment issues</i>	Sludge management (quantity, stability)	Limited knowledge	Low kinetics of N oxidation	
Research work (RIME)	<b>Use of natural and residual materials (raw vs modified) for prevention and control of mine water contamination, and mine sites rehabilitation</b>			



## Natural and residual materials for mine water treatment and sites rehabilitation

Case study	AMD/NMD prevention	Passive treatment
<i>I: East-Sullivan mine site</i>	Residual organics cover	Constructed wetlands + water pumping through the organic cover
<i>II: Manitou mine site</i>	Desulfurized non-acid generating tailings cover	(To be designed and constructed)
<i>III: Wood-Cadillac mine site</i>	Inert sand-based cover	Wood-based biofilter
<i>IV: Lorraine mine site</i>	CCBE (cover with capillary barrier effect) – multi-layer	Anoxic dolomite drains + tri-unit biochemical train
V: New materials	Modification / Improvement	Use
<i>Charred dolomite</i>	Enhanced specific surface and porosity, increased pH and alkalinity generation	Synthetic NMD treatment
<i>Modified wood ash</i>		Real NMD treatment
<i>Activated biochar</i>	Porosity arrangement	Real AMD treatment
<i>N-rich residuals</i>	N/A	Non-acid generating tailings revegetation

- Ongoing research
- Concluding remarks

## Location of East-Sullivan and Manitou mine sites



Val d'Or town

East-Sullivan  
mine site

Manitou  
mine site

# East-Sullivan mine site: operation, abandonment, rehabilitation

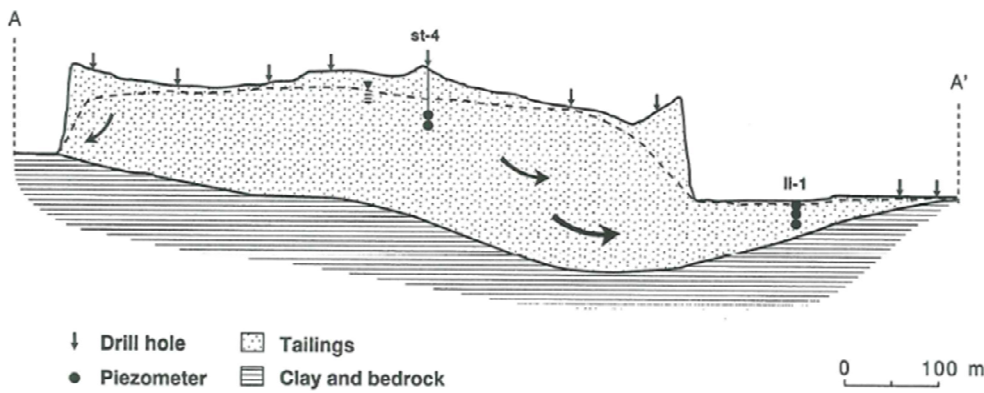


Figure 2. Cross-section A-A' through the East-Sullivan tailings impoundment. Arrows indicate locations of drill holes to the bedrock or clay basement. Dots represent piezometer locations for water sampling.



## Low pore-water quality in 1992

- pH  $\approx$  2
- Fe ( $\text{Fe}^{2+}$ ): up to 17 g/L
- $\text{SO}_4^{2-}$ : up to 37 g/L
- Cu, Pb, Zn : 0.1-1 g/L

➔ High strength AMD

Long-term treatment OR prevention and temporary polishing?

(<http://sebastienlavoie.com/maitrise/photos.html>; <http://www.mrn.gouv.qc.ca/mines/restauration/restauration-sites-east-sullivan.jsp>; <https://www.oiseauxduquebec.org>)



## East-Sullivan organic cover: mine site rehabilitation

- **1984: Organic waste (residual wood and biosolids) cover instalment for AMD prevention and [temporary] treatment**
- **1990:** Seepage collection system
- **1992-1996:** Confining dike (6 km) + water polishing in constructed wetlands
- **1998-2005:** Collected AMD in constructed wetlands pumping through the organic cover
- **2019-2020:** Wood cover completion  
⇒ **Some effluents were still acidic**

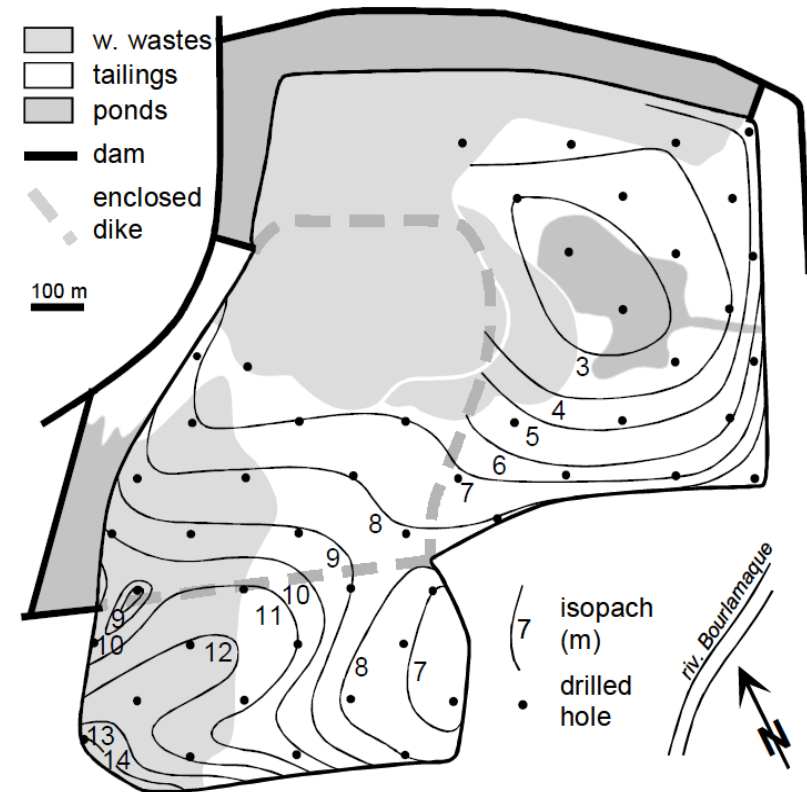
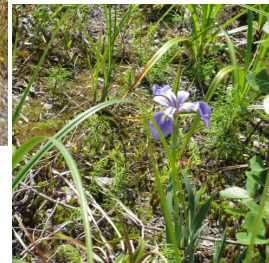
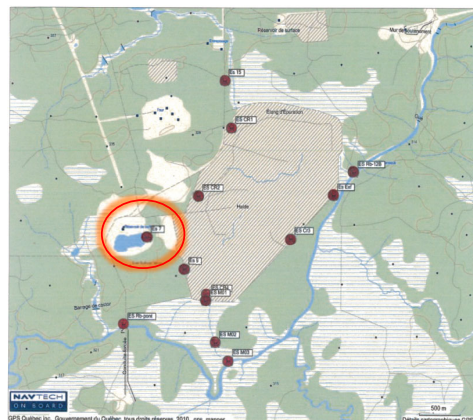


Figure 1. Map of the East Sullivan tailings impoundment in 1994

# East-Sullivan organic cover: mine site monitoring

- **Network of sampling points**
  - over 20-year data
- **Parameters analyzed**
  - pH, Fe, Cu, Zn (+ Al, Mn, Pb, SO<sub>4</sub><sup>2-</sup>)
- **Compliance, except for the last covered tailings**



- Covered tailings and constructed wetlands: **blooming vegetation** and **birds' refuge** (> 190 species listed)



# Desulfurized tailings cover: Manitou mine

- Prevention and pretreatment of AMD (pH: 2-3; 10-12 g/L Fe; 0.6-1 g/L Zn; 0.1-1 g/L Cu; 30-40 g/L  $SO_4^{2-}$ )

Production site (Goldex)

Abandoned mine site (Manitou)

20 km

No need for a new tailings storage facility (↓ environmental footprint)

© GoogleMaps

Reclamation of an abandoned mine site

Efficient to prevent AMD generation (+ additional performance ↗ because of residual sulfides)

2007

Instrumentation and monitoring

Slurry deposition

© RIME

© RIME

2017

Cover materials ≠ borrowed in the environment

© MERN

© Agnico Eagle Mine

Leachate still contaminated because of pre-oxidation  
→ Passive treatment system (to be designed and constructed, as AMD is now pretreated to NMD)



# Sand cover + wood-based biofilter: Wood-Cadillac mine site

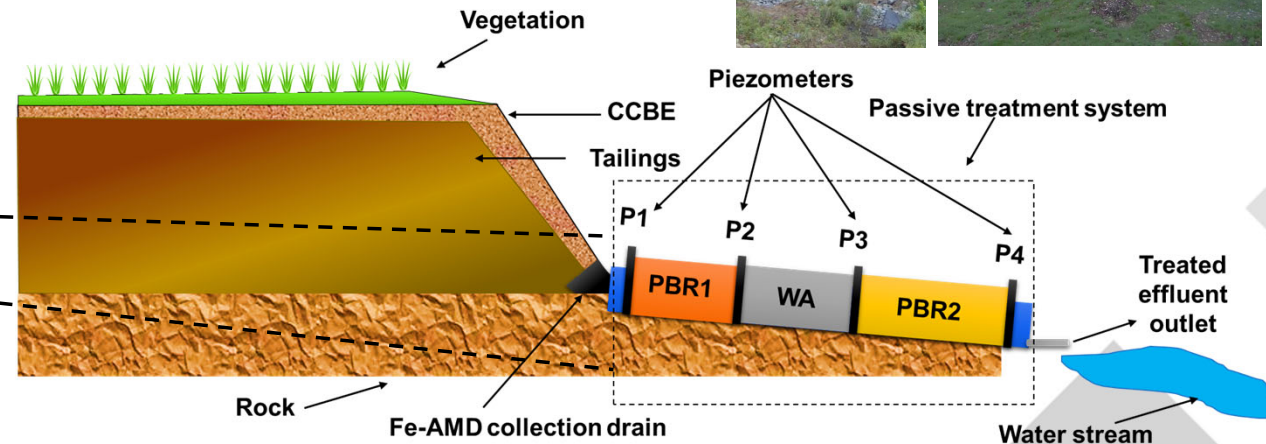
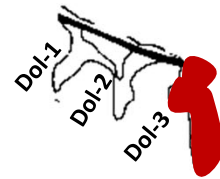
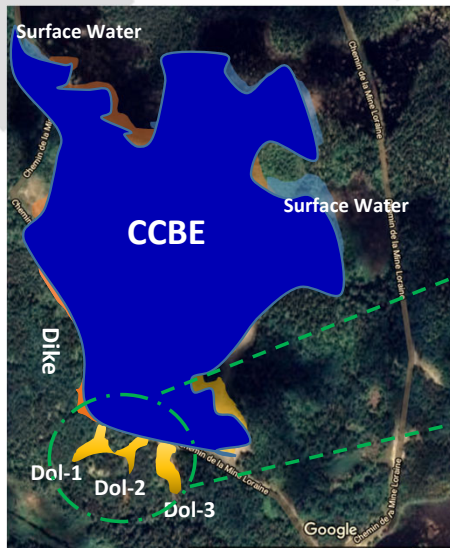
- Efficient passive treatment of As-NMD: removal of As & metals; decrease of  $\text{SO}_4^{2-}$  [ ] to  $< 200 \text{ mg/L}$

The collage consists of several panels:

- Map:** Shows Québec with a red box labeled 'A.T.' and 'Rouyn-Noranda'. A dashed line indicates the location of the mine site near 'Amos' and 'Road 117', with 'City of Cadillac' nearby. Credits: © Wikipedia, © GoogleMaps.
- Abandoned mine site (Wood-Cadillac):** Aerial view of the site. Credit: © RIME.
- Reclaimed mine site (environmental footprint):** Aerial view of the site after reclamation, outlined in red. Credit: © RIME.
- Biofilter:** Aerial view of the biofilter structure. Credit: © RIME.
- Reclamation of an abandoned mine site:** A vertical timeline of four photos showing the site's progress from 1995 to 2022. Credits: © MERN.
- Sampling of poral waters:** Photo of a sampling station on a boat. Credit: © RIME.
- In-situ poral water monitoring:** Photo of a water sample in a bottle. Credit: © RIME.
- Fresh residues (wood chips):** Photo of a container filled with wood chips. Credit: © RIME.
- Residues characterization:** Photo of a laboratory setup for residue analysis. Credit: © RIME.
- Passive biofilter:** Aerial view of the biofilter in 2022. Credit: © MERN.

(Germain & Cyr, 2003; Libéro, 2007; Mehdaoui et al., 2024; Thevenot et al., 2024)

# CCBE + passive treatment: Lorraine mine site



- **1964-1968:** Extraction of **Cu, Au, Ag, and Ni**

- Acid generating tailings: 15.5 ha (up to 6 m)
- AMD: pH=3.6, 7 g/L Fe and 15 g/L sulfates

- **In 1998: Mine site reclamation**

- Multi-layer dry cover with capillary barrier effect (**CCBE**): O<sub>2</sub> prevention
- **AMD treatment: 3 anoxic dolomite drains** (Dol-1 to Dol-3)

- **In 2011: Dol-3 clogged**, replaced by **tri-unit system: PBR1-WA-PBR2**

(AMD: pH < 4, 3g/L Fe)

- **PBR1:** 40% organics, 60% inorganics (pH ↑, sulfate removal)
- **WA:** 100 % wood ash (**Fe treatment**)
- **PBR2:** 77% organics, 23 % inorganics (**polishing**)





- Tri-unit system **progressive loss of efficiency**: PBR1-WA-PBR2

- **Porosity clogging** by Fe minerals
- **Preferential flow** and partial water **bypassing** the system

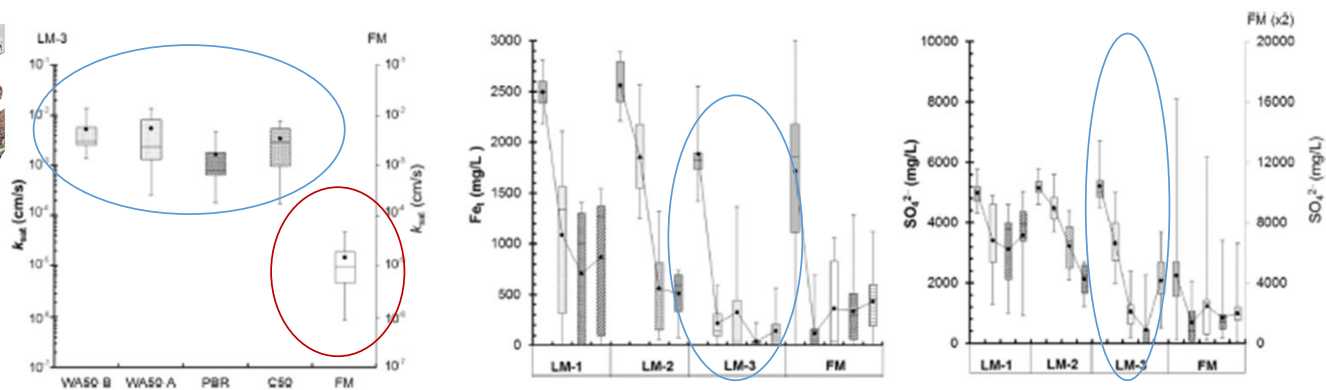
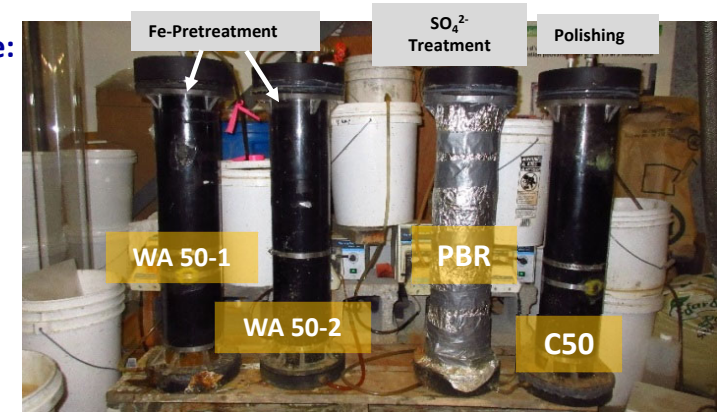


**New multi-step system ?**

- Multi-step system **calcite-/dolomite-DAS (dispersed alkaline substrate: coarse material + neutralizing agents) & passive biochemical reactors**

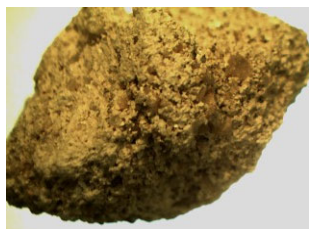
- 2 Fe pretreatment units (50% wood chips + 50% wood ash)
- 1  $\text{SO}_4^{2-}$  treatment unit (70% organic + 30% inorganic matter)
- 1 polishing unit (50% calcite + 50% wood chips)

- **Correction factor of 0.2–0.8 for Q and Fe removal + factor 1/100 for  $k_{\text{sat}}$**  (measured in laboratory)





## New materials: Sources and modification procedures



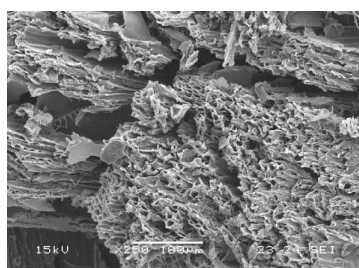
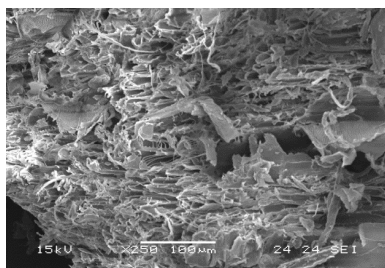
Material	Parameter		Composition (%)		
	pH <sub>paste</sub>	Porosity	Dolomite [CaMg(CO <sub>3</sub> ) <sub>2</sub> ]	Calcite [CaCO <sub>3</sub> ]	Magnesia [MgO]
Raw dolomite	7.9	0.44	<b>87.1</b>	BDL	BDL
<b>Half-charred dolomite</b>	<b>11.6</b>	<b>0.56</b>	<b>7.2</b>	<b>53.7</b>	<b>19.9</b>

- **Half-charred dolomite: dolomite content decreased, two new minerals were created**

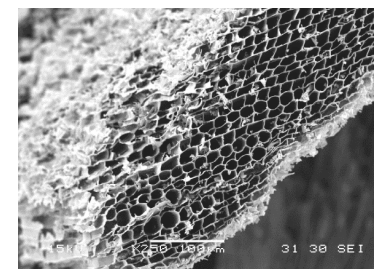


Parameter	Ash B	<b>Ash B modified</b>	Ash W	<b>Ash W modified</b>
pH <sub>paste</sub>	13.8	<b>12.6</b>	9.3	<b>12.8</b>
CEC, meq /100g dry	138	<b>322</b>	66	<b>311</b>

- **Wood ash: modification generated high CEC and paste pH new materials**



**Step 2:** Activation  
(chemically: KOH, H<sub>3</sub>PO<sub>4</sub> or  
physically: steam, CO<sub>2</sub>)

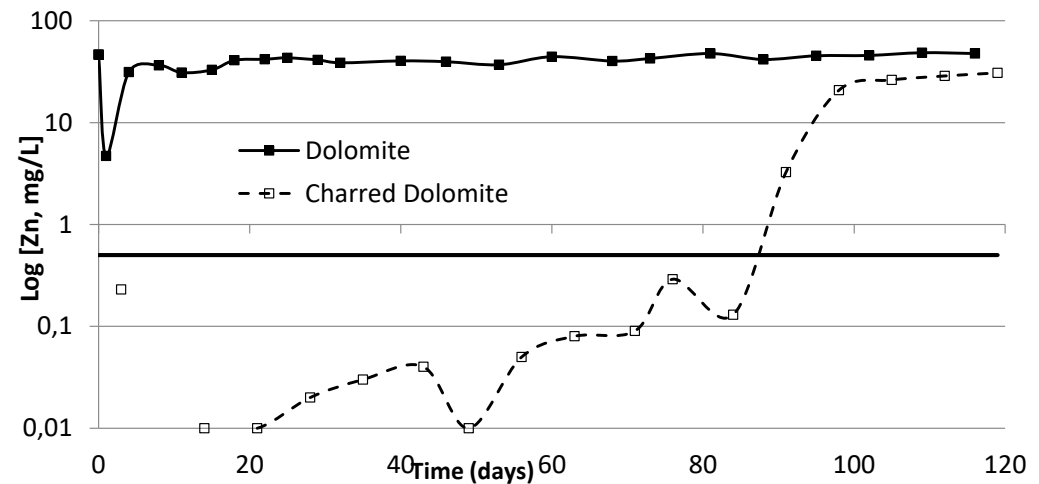
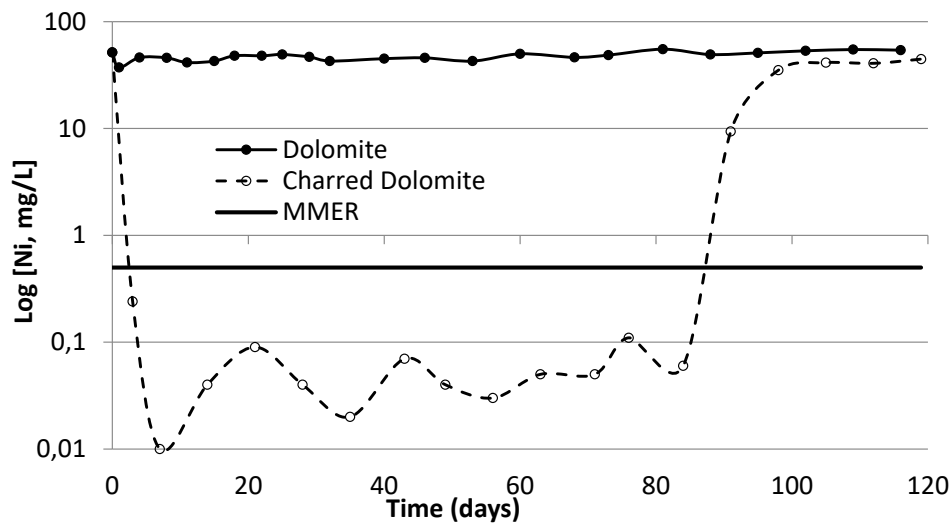


- **Activated biochar: arranged porosity**

**Step 1:** Torrefaction, slow to flash pyrolysis, or gasification under different operating conditions

# Raw vs modified dolomite: Ni, Zn removal in synthetic NMD

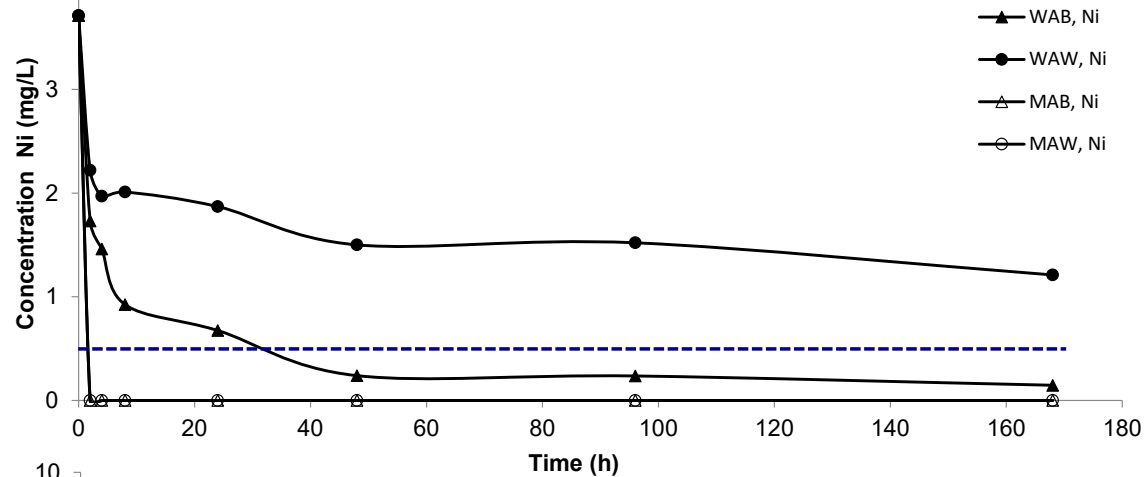
- Significantly better efficiency of charred dolomite for Ni and Zn removal (50 mg/L each)



Criteria

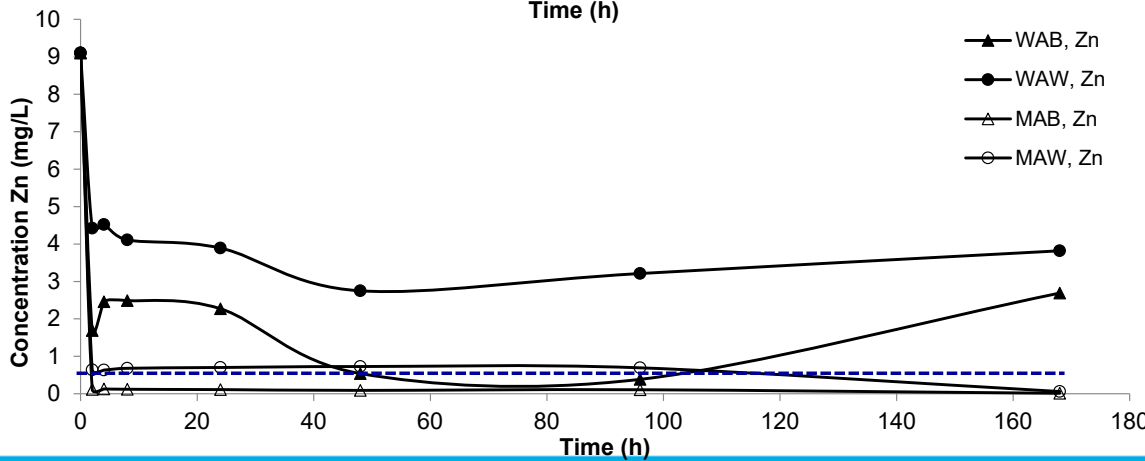
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# Raw vs modified wood ash: Ni, Zn removal in real NMD



## Effluent #1 (pH 7.89, 3.71 mg/L Ni)

- Ni removal (<0.5 mg/L)
  - 2h for both modified ash (MAB & MAW)



## Effluent #2 (pH 6.85, 9.1 mg/L Zn)

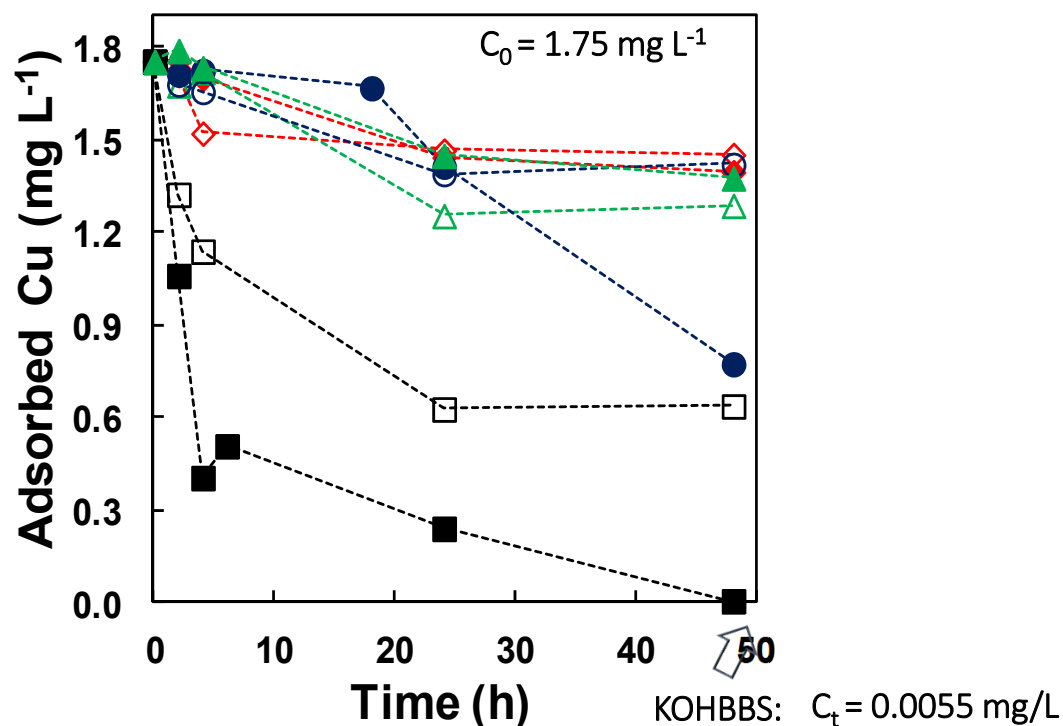
- Zn removal (<0.5 mg/L)
  - 2h for MAB
  - 7 days for MAW (93% within 2h), but 2h for Mn removal (99% of 4.2 mg/L)



## Activated biochar: Cu removal in real AMD

- **KOHBBS**: Efficient for Cu removal in **real** effluents

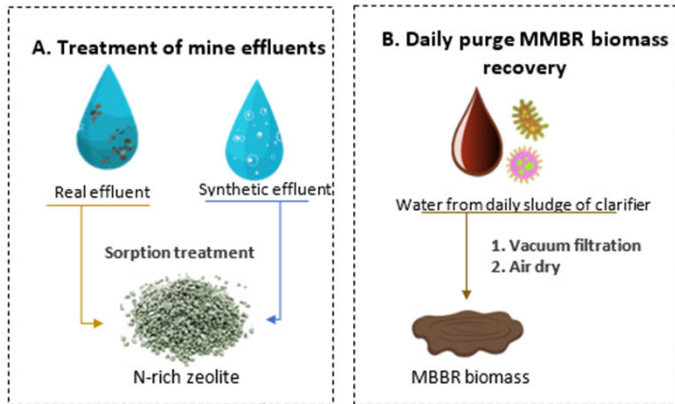
$S_{\text{BET}} = 1700 \text{ m}^2/\text{g}$ ; 100% de micropores; 22.4% oxygenated groups



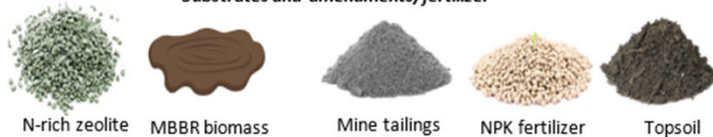
Parameter	Real AMD (mg/L)	After adsorption (KOHBS) (mg/L)	Efficiency (%)
Co	9.4	0.5	95 ↓
<b>Cu</b>	<b>1.75</b>	<b>0.006</b>	<b>~ 100 ↓</b>
Fe	468	405	13 ↓
Mn	10.9	9.7	11 ↓
Pb	0.14	0.08	43 ↓
Zn	4.9	4.6	6 ↓

# N-rich residuals use in tailings revegetalization

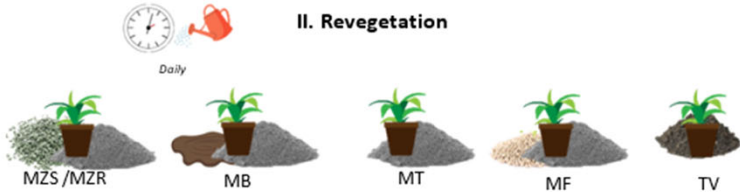
## I. Treatment of $\text{NH}_3\text{-N}$



### Substrates and amendments/fertilizer



## II. Revegetation



1. Above-ground biomass and root biomass
2. Physicochemical characterization of the surface runoff at 14, 28 and 90 days
3. Foliar analyses for nutrient content and phytoaccumulation of trace elements

## Summary of main findings

### N-rich zeolite

- Plant biomasses like tailings alone
- **Foliar Na** concentrations 6-9 times higher than in other treatments

vs

### MMBR Biomass

- Plant biomasses like fertilized tailings and topsoil
- **High Se** concentrations in leaves

**Better performance**

- Foliar N concentrations and root biomasses failed to discriminate between the two tested types of amendment



# Ongoing research

## Scientific knowledge for informed new practical applications

- **Organic cover:** is elevated water table required? Is water pumping through the cover necessary?
- **Low-sulfides no-acid generating tailings cover:** evolution and fate of potential contaminants under oxic vs anoxic and abiotic vs biotic conditions
- **Passive NMD treatment in residual organics-based biofilters:** contaminants removal mechanisms and residues stability\*
- **Raw vs half-charred dolomite:** prevention of AMD generation from pyrrhotite-rich tailings and passive polishing
- **N-rich residuals from mine water treatment:** potential of surrounding environment contamination (uptake by vegetation, runoff)

\*Some findings to be presented during ICARD 2024 (Mehdaoui et al., 2024; Thevenot et al., 2024)

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## Concluding remarks

- **Successful rehabilitation approach** for oxidized tailings on mine sites (precious and base metals) **must combine prevention** (tailings covering) and **passive treatment**
- **Residual materials valorization** (already available on or in the proximity) limits disposal concerns, environmental footprint, and mine sites rehabilitation costs
- **Materials** (natural and residual): efficient in **mine tailings covers** for AMD prevention or **transformed**, with promising results in contaminated mine water treatment
- Metal recovery, whenever feasible, **could decrease water treatment costs**
- Pilot scale production and testing of modified materials is limited
- Metal recovery, sorbent and treated water reuse, are rarely addressed

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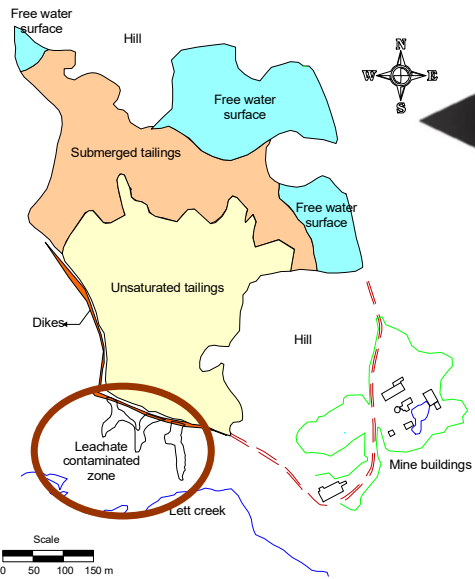
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Merci!



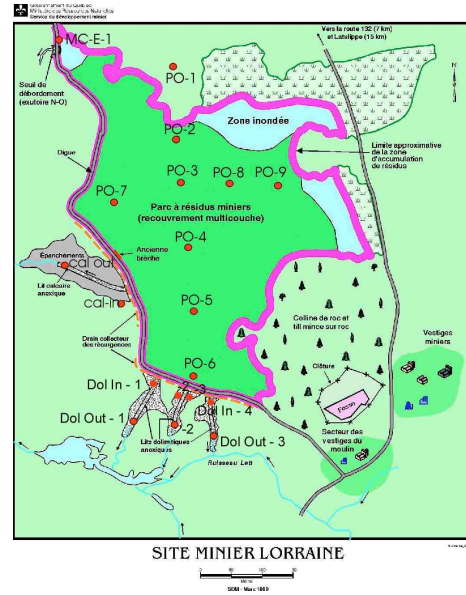




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# CCBE + passive treatment: Lorraine mine site

2000: CCBE + 3 dolomitic drains

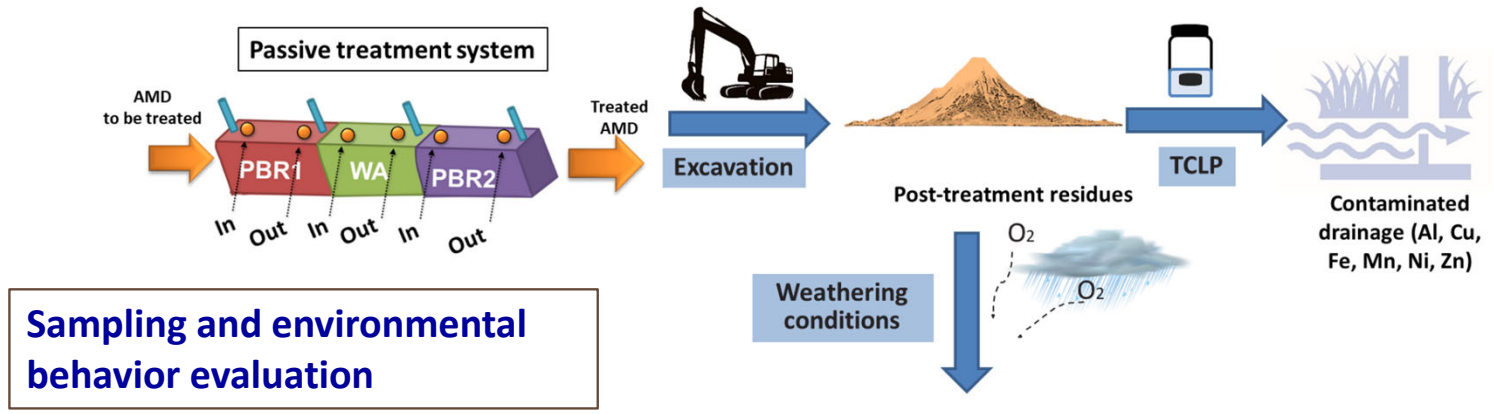


1964-1968 : extraction of Cu, Au, Ag, Ni  
acid-generating tailings: 15.5 ha (up to 6 m)

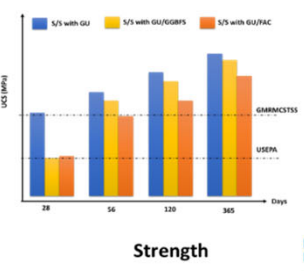
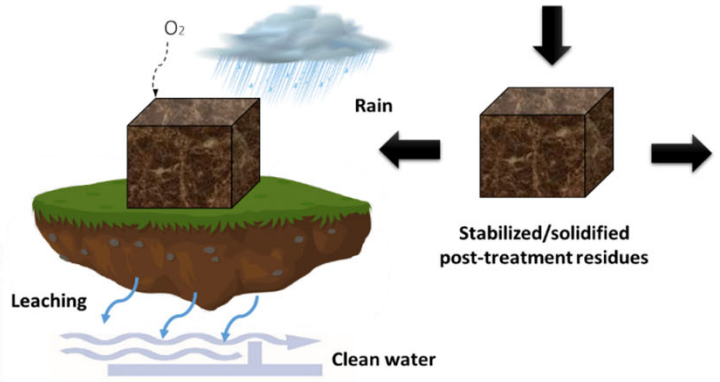
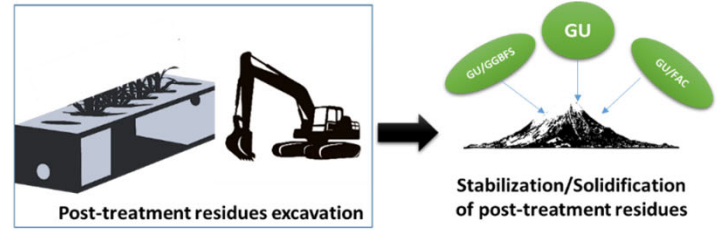
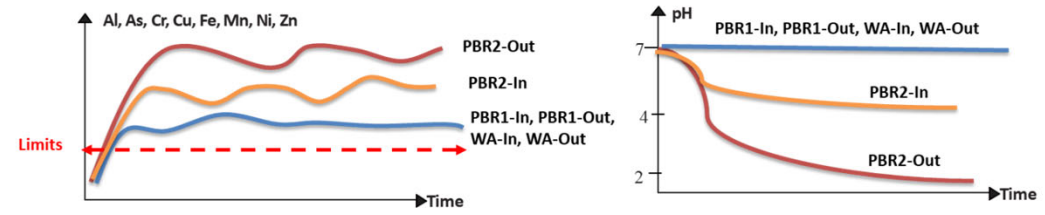


AMD: pH 3.6, 7 g/L Fe, 15 g/L sulfate

# Geochemical stability of AMD treatment solids



**Sampling and environmental behavior evaluation**



**Stabilization**



# Plant material and growing conditions



<sup>1</sup>(Guitttony, 2021; Tordoff et al., 2000)

<sup>2</sup>(Guitttony, 2021)

<sup>3</sup>(Wiggans et Frey, 1955)

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