

# **THE ROTATING CYLINDER TREATMENT SYSTEM RCTS**

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# **Treatment of Acid Mine Drainage (Lime Precipitation)**

## **-The addition of lime:**

- 1. to raise the pH and precipitate metals as hydroxides**
- 2. to precipitate sulfate as gypsum**

## **-Requires:**

- 1. oxygen addition if dissolved ferrous iron is present**

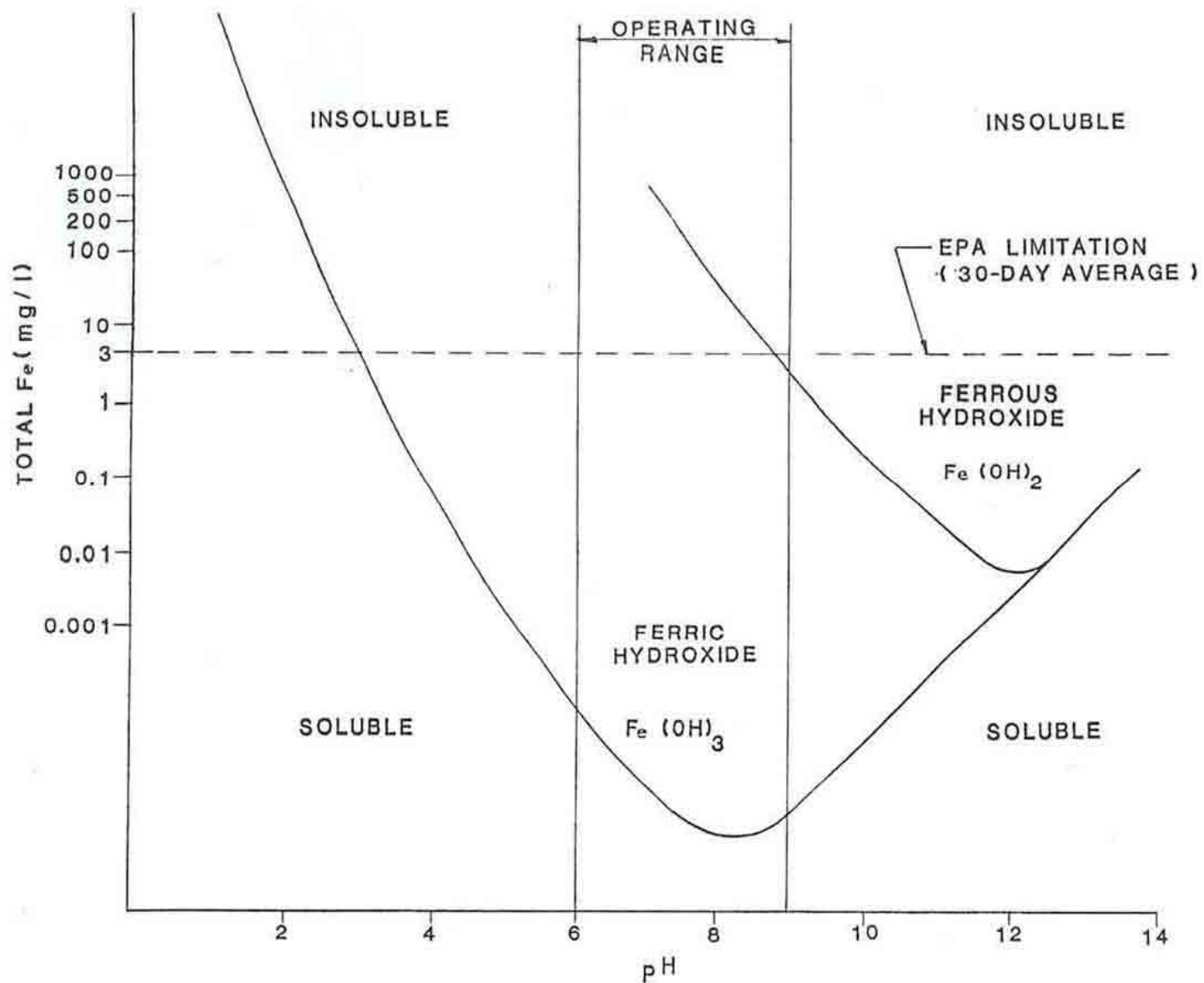
**Oxygen addition is typically accomplished with compressors and air diffusers placed in reaction tanks.**

- 2. thorough mixing due to its low solubility and slow dissolution rate.**

**Mixing is typically accomplished with mixers inside of reaction tanks.**

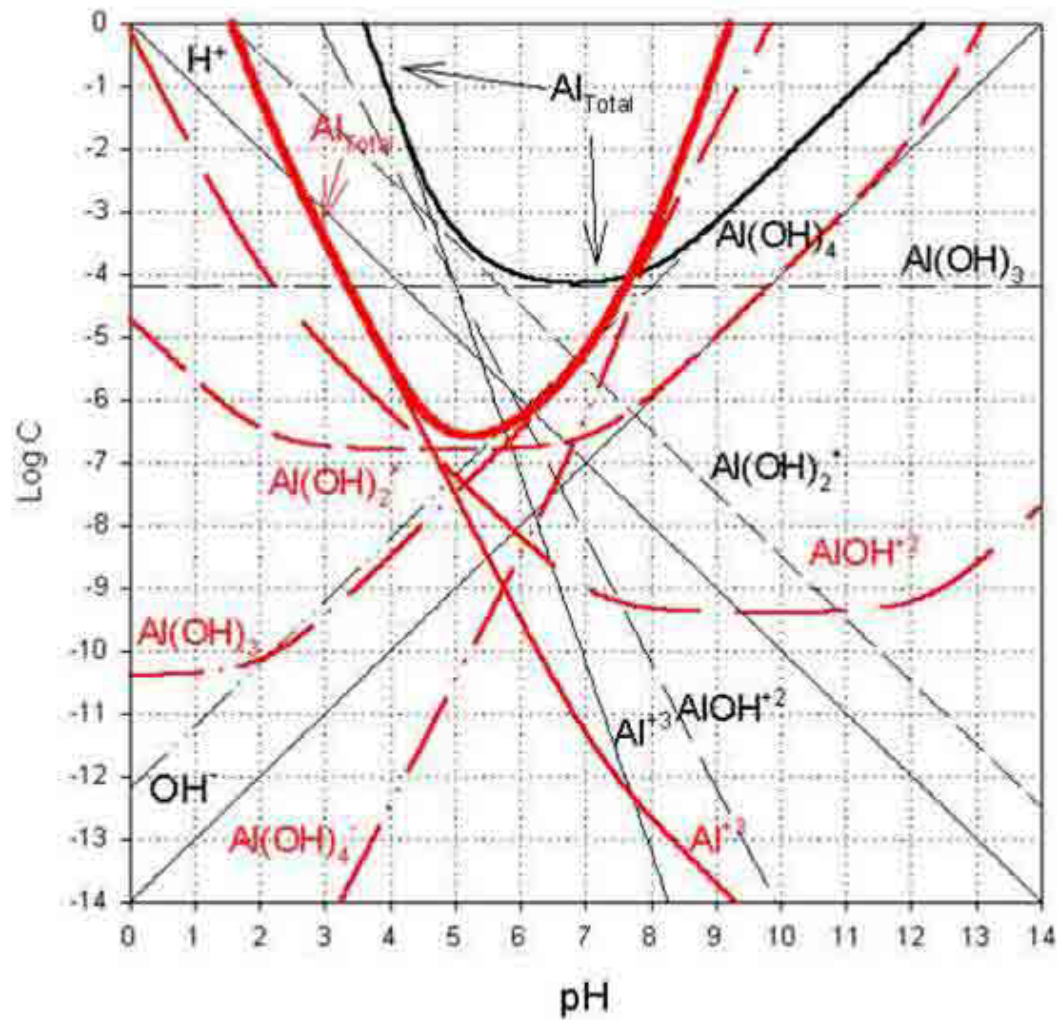
**-Typically labor intensive due to the requirements listed above**

# Iron Hydroxide Solubility vs pH

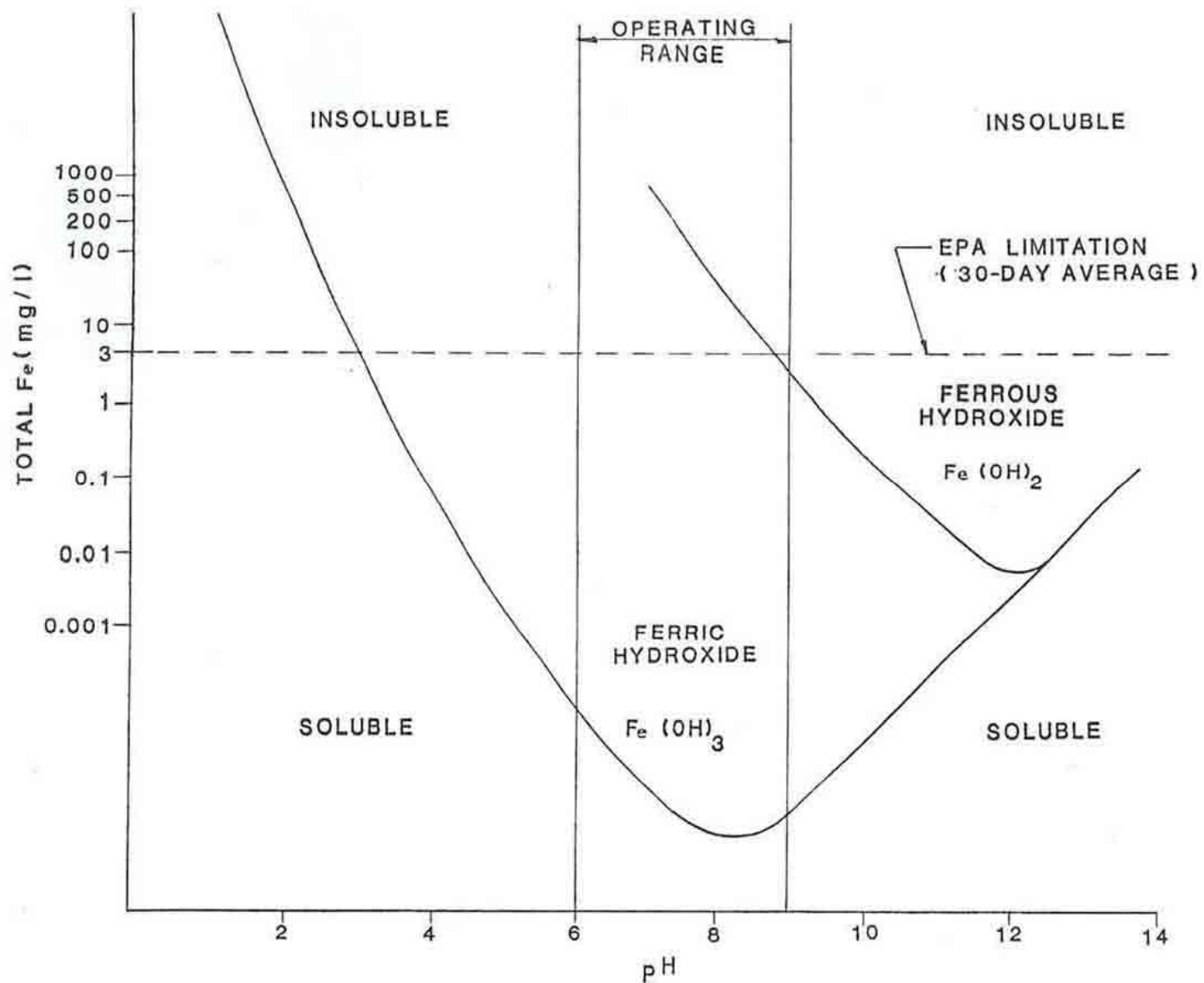




# Aluminum Hydroxide Solubility vs pH



# Iron Hydroxide Solubility vs pH



# **DESIGN AND OPERATIONAL CONCEPT OF THE RCTS TREATMENT SYSTEM**

- **Rotating perforated cylinders add oxygen from the atmosphere to the water**
- **Compressors and blowers are eliminated**
- **Aggressive agitation maximizes reagent efficiency**



# Improved Oxygen Addition

**Provides more oxygen per energy consumed than conventional systems**

**mechanical surface aeration systems provides 3.0-3.5 lbs of oxygen per horsepower hour (USEPA 1983)**

**submerged turbine aerators utilizing dual impeller turbines provide 2.5-3.0 pounds of oxygen per horsepower hour (USEPA 1983)**

**600 gallon four rotor RCTS provided approximately 9 pounds of oxygen per horsepower hour**

$$O_2 = Q_w \times Fe \times 7.14 \times 10^{-5}$$

**$O_2$  = Theoretical  $O_2$  demand (lb  $O_2$ /hr)**

**$Q_w$  = Acid mine drainage flow rate (gal/min)**

**$Fe$  =  $Fe^{2+}$  initial concentration (mg/L)**

**Operated on two cylinders powered by 0.375 hp and oxidized ~ 5,000 mg/L of iron at a flow rate of 10 gallons per minute.**



# Four Rotor RCTS Unit









# Single Rotor High Speed RCTS Unit (RCTS-HS)





# **RCTS TECHNOLOGY AT THE RIO TINTO MINE IN NORTHEASTERN NEVADA 2003**



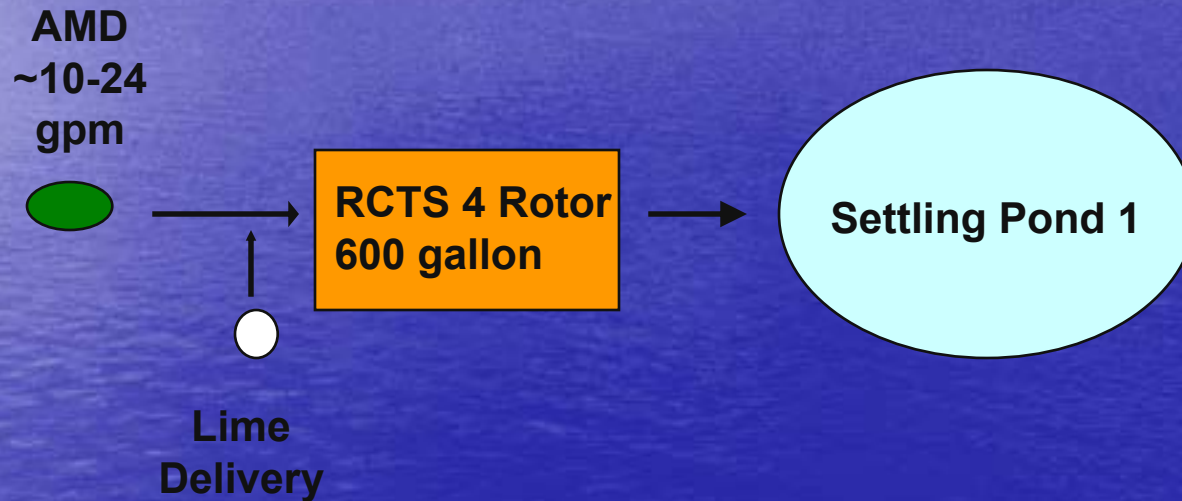
# **THE RIO TINTO MINE**

## **Highly concentrated AMD**

- **The RCTS treated AMD with  $\text{Fe}^{2+}$  concentrations approaching 4,900 mg/l.**
- **Acidity was in excess of 12,500 mg/l.**
- **Sulfate concentrations were in excess of 18,000 mg/l.**



# Treatment Schematic 2003 RCTS 4 Rotor Rio Tinto Mine



# RIO TINTO MINE IN 2003

- **600 Gallon Prototype Unit met all Water Quality Standards applicable at the site**
- **Specific experiments undertaken for the Hybrid RCTS-Sulfate Reducing System**
- **Aluminum concentrations of 546 mg/l were removed to .009 mg/l during this experiment by the RCTS**





# Treatment Results 2003 RCTS 4 Rotor Rio Tinto Mine

Table 2.4.1. 2003 RCTS Results

## Dissolved Influent Concentrations

Date	TDS	Al	As	Cd	Cu	Cr	Fe	Mn	Se	Zn
9/19//03	6510	124	0.0066	0.0964	53.7	0.0126	833	27.4	0.23	16.5
9/20/03	6960	133	0.0077	0.0968	57.9	0.0134	931	29.4	<0.05U	17.9
9/21/03	24200	528	0.024	0.338	215	0.044	4640	86.5	<0.2U	62.1
9/22/03	23300	540	0.026	0.342	220	0.044	4790	89.1	<0.2U	64.7
9/22/03	25200	546	0.026	0.340	222	0.044	4870	90.8	<0.2U	65.2

## Dissolved Effluent Concentrations

	TDS	Al	As	Cd	Cu	Cr	Fe	Mn	Se	Zn
9/19//03	4240	0.246	<0.005U	0.0006	0.016	0.0006	<0.05	3.54	<0.001U	0.06
9/20/03	4500	0.247	<0.005U	0.0007	0.017	0.0015	<0.05	4.18	<0.001U	<0.05
9/21/03	8780	0.109	<0.005U	<0.0005	0.019	0.0019	<0.05	3.35	0.001	0.05
9/22/03	7380	0.077	<0.005U	<0.0005	0.020	0.0019	<0.05	2.57	<0.001U	<0.05
9/22/03	12300	0.009	<0.001U	0.025	0.093	0.004	<0.10	52.2	<0.001U	0.70
9/24/03	9810	0.068	<0.002U	<0.0005	0.010	<0.002U	<0.10	3.46	0.001	<0.05
9/24/03	9780	0.071	<0.002U	<0.0005	<0.010	<0.002U	<0.10	3.46	0.001	<0.05

# THE RIO TINTO MINE 2004

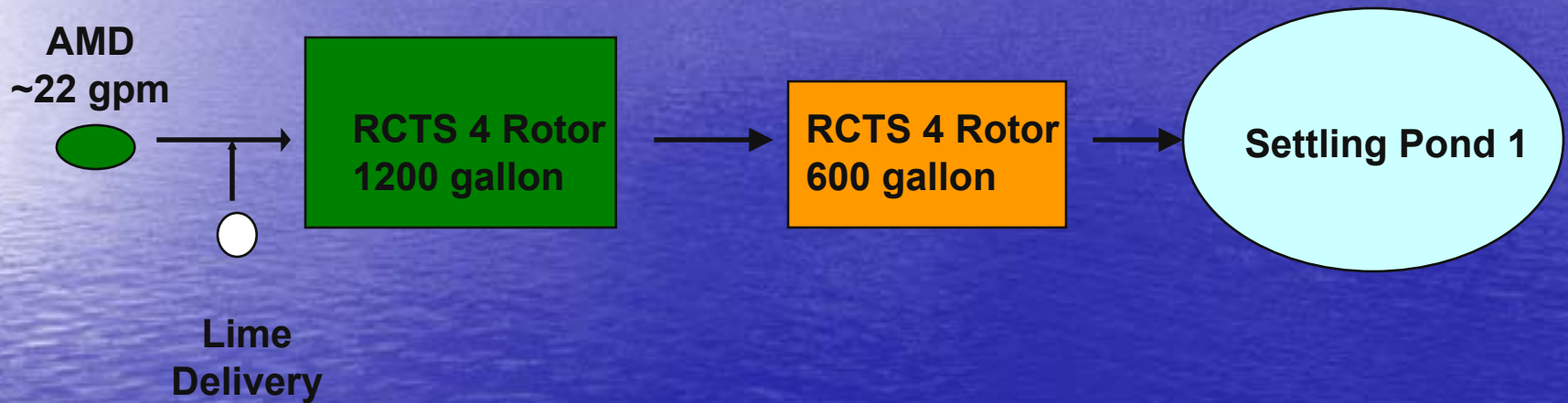




# THE RIO TINTO MINE (4-ROTOR) TREATMENT 2004



# Treatment Schematic 2004 RCTS 4 Rotor x 2





# **THE RIO TINTO MINE**

## **RCTS-HS 2004**





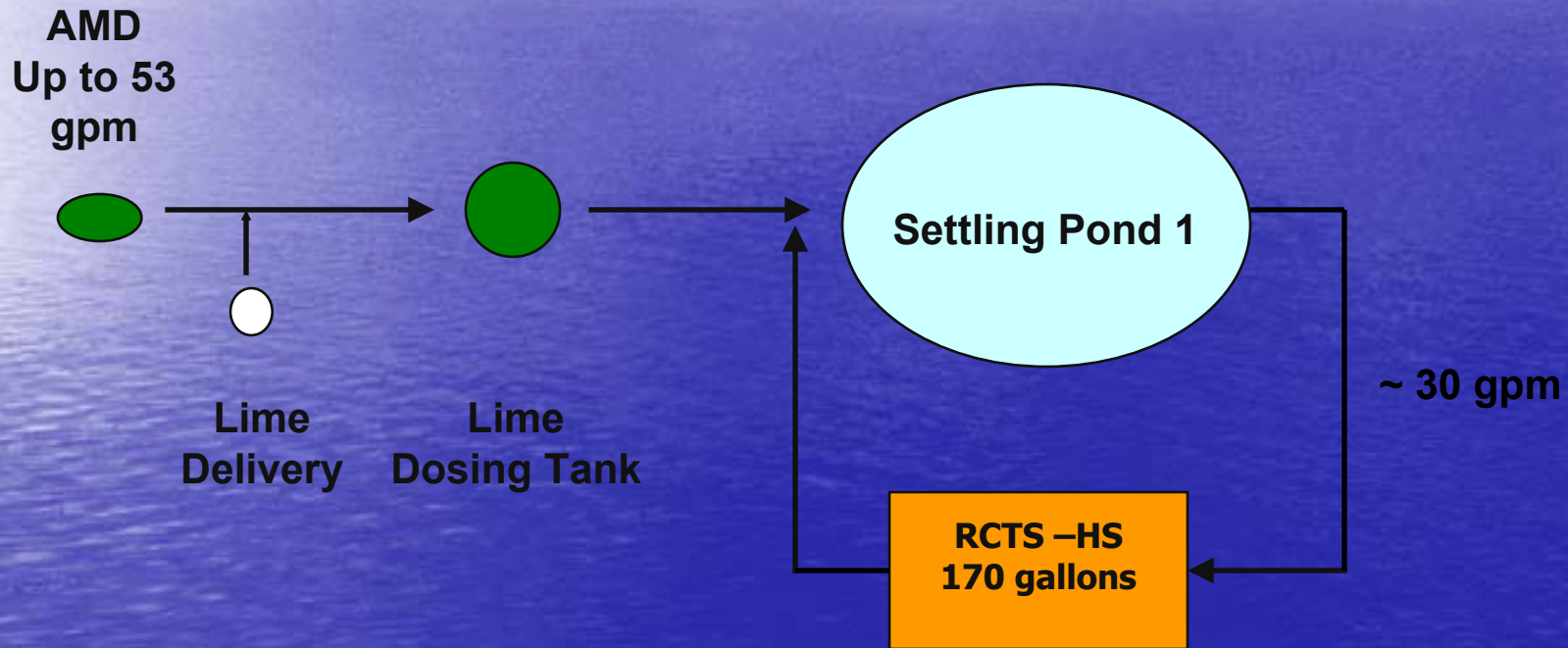
# RCTS-HS 2004

- The RCTS-HS prototype was utilized for an emergency water level adjustment of a hydraulic control pond in November of 2004.





# Treatment Schematic 2004 RCTS –HS Rio Tinto Mine



# TYPICAL RESULTS

mg/l

INFLUENT	Unit Type	Al	As	Cd	Cu	Fe	Mn	Zn
10/4/04	RCTS 4 Rotor x 2	491	nd	.32	202	4170	77.5	54.6
10/6/04	RCTS 4 Rotor x 2	433	.03	.278	209	3550	82.6	50.9
11/9/04	RCTS-HS	683	.03	.365	293	4800	109	74.3

EFFLUENT	Unit Type	Al	As	Cd	Cu	Fe	Mn	Zn
10/4/04	RCTS 4 Rotor x 2	nd	nd	nd	.037	.74	4.51	.07
10/6/04	RCTS 4 Rotor x 2	nd	nd	.002	.097	.11	9.54	.09
11/9/04	RCTS-HS	nd	nd	nd	.003	nd	4.88	nd



# **RCTS TECHNOLOGY AT THE RIO TINTO MINE IN NORTHEASTERN NEVADA 2004**

## **Results:**

- **The RCTS met all Federal Water Quality Standards applicable at the site in a single stage pH adjustment of the influent.**
- **The RCTS-HS met all Federal Water Quality Standards applicable during this emergency lagoon type treatment.**
- **Treated ~22 gallons/min (RCTS 4 rotor x 2)  
~53 gallons/min (RCTS-HS)**
- **Operated on less than 1600 watts of electricity.**
- **Lime slurry efficiency ~98%**



# RIO TINTO MINE

## 2005





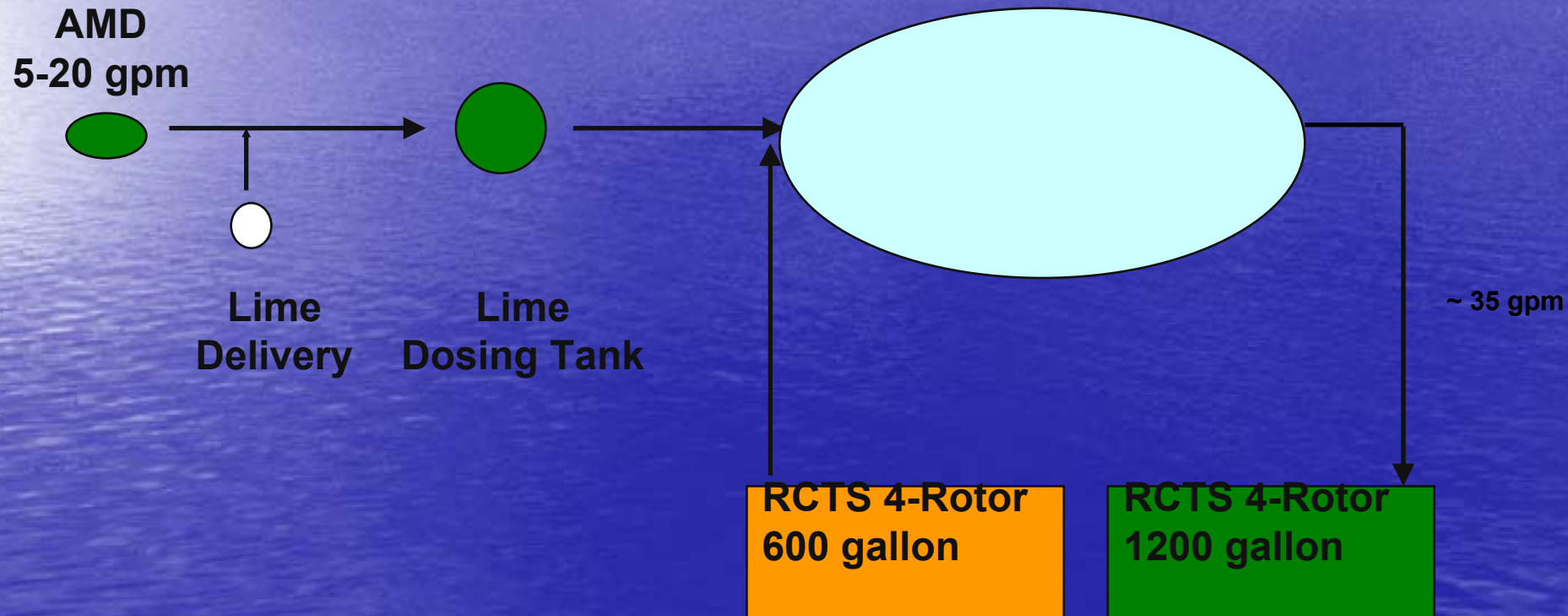
# **RIO TINTO MINE**

## **2005**



# RIO TINTO MINE

## 2005





# TYPICAL RESULTS

## mg/l

**Table 1. Treatment Results for the Rio Tinto Mine 2005 Concentrations (mg/L)**

Date	Sample location	Al	As	Cd	Cu	Fe	Mn	Zn	Sulfate
7/19/2005	Influent	726	nd	0.340	320	6780	87.3	73.6	24,100
7/19/2005	Effluent	0.2	nd	nd	0.005	nd	1.01	nd	4110
7/26/2005	Influent	793	0.03	0.359	314	6890	96	79.4	24,180
7/26/2005	Effluent	0.1	nd	0.0005	0.005	nd	0.52	nd	2,410
8/5/2005	Influent	540	nd	0.338	228	4990	80.3	60	17,600
8/5/2005	Effluent	0.08	nd	0.0002	0.002	0.05	0.41	nd	1,800
8/11/2005	Influent	297	nd	0.210	130	2840	63.9	36.7	10,200
8/11/2005	Effluent	0.13	nd	nd	0.01	nd	0.2	nd	1,950
8/18/2005	Influent	305	nd	0.200	128	2950	58.1	35.2	10,900
8/18/2005	Effluent	0.114	nd	nd	0.014	0.06	0.2	nd	2,070
9/7/2005	Influent	572	nd	0.301	248	5110	67.4	57.5	17,600
9/7/2005	Effluent	0.26	nd	nd	0.018	0.40	0.23	nd	2,560
9/23/2005	Influent	325	nd	0.198	139	2940	58.2	36.5	9,710
9/23/2005	Effluent	0.07	0.001	0.0002	0.009	nd	0.58	nd	2,390
9/30/2005	Influent	279	nd	0.230	131	2570	51.9	32.3	9,910
9/30/2005	Effluent	0.04	0.001	0.0002	0.011	nd	0.581	nd	2350

# RCTS TECHNOLOGY AT THE LEVIATHAN MINE 2004

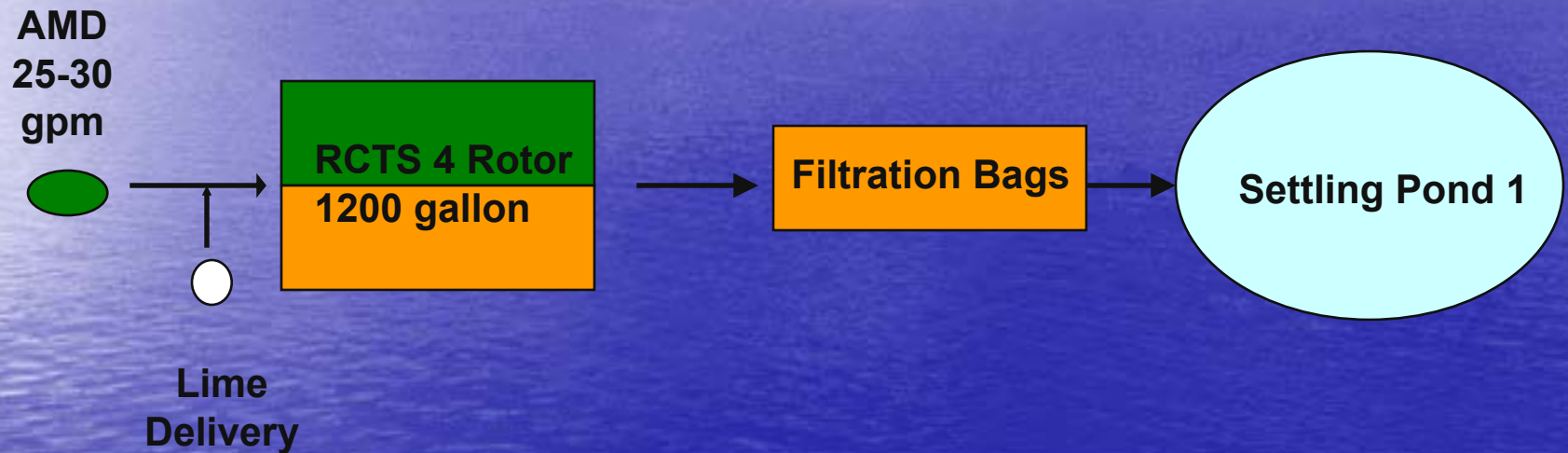




# **RCTS TECHNOLOGY AT THE LEVIATHAN MINE 2004**



# Treatment Schematic 2004 RCTS 4 Rotor Leviathan Mine





## **LEVIATHAN MINE 2004**

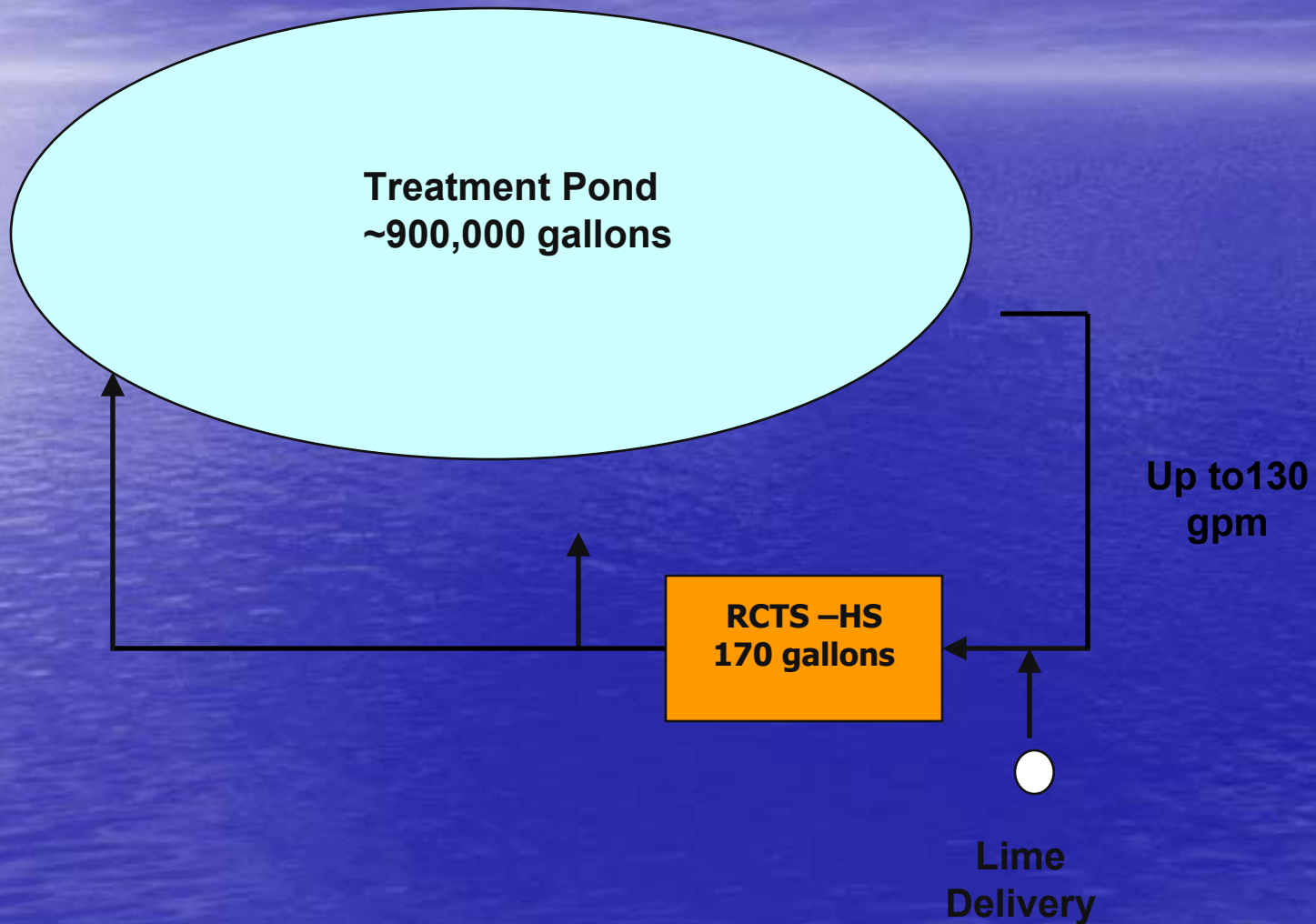
- **The RCTS treated 25-30 gpm Fe<sup>2</sup> concentrations of 300 to 400 mg/l.**
- **Lime slurry efficiency was 41% better than the conventional tank reactor system onsite.**
- **Influent residence time within the RCTS was 75% less than the conventional tank reactor system onsite.**
- **Water Quality Standards were met by the RCTS in a single stage pH adjustment.**
- **The RCTS operated on less than 1600 watts of electricity during treatment operations.**

# **RCTS TECHNOLOGY AT THE LEVIATHAN MINE EMERGENCY BATCH TREATMENT 2005**





# Treatment Schematic 2004 RCTS –HS Northeastern Nevada



# **RCTS TECHNOLOGY AT THE LEVIATHAN MINE EMERGENCY BATCH TREATMENT 2005**

- **Mobilize the system in 3 days**
- **The RCTS treated ~800,000 gallons of AMD in approximately 90 hours.**
- **Acidity ~ 1,300 mg/L mostly aluminum**
- **All applicable Water Quality Standards were met**
- **The RCTS operated on less than 1600 watts of electricity during treatment operations**
- **Lime efficiency > 89%**





# **RCTS TECHNOLOGY AT THE LEVIATHAN MINE EMERGENCY BATCH TREATMENT 2005**





# **RCTS TECHNOLOGY AT THE LEVIATHAN MINE EMERGENCY BATCH TREATMENT 2006**





# EMPIRE MINE GRASS VALLEY CALIFORNIA



**Flow ~ 6 gpm**

**pH ~ 6.6**

**iron ~ 4 mg/L**

**arsenic ~ 0.05 mg/L**



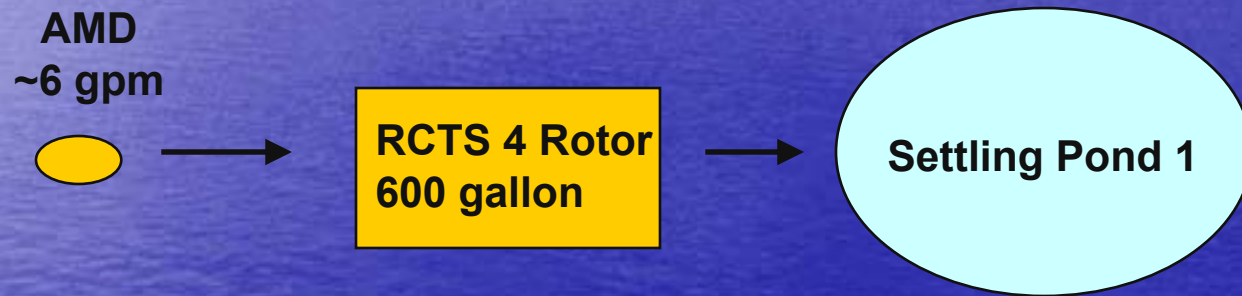
# EMPIRE MINE GRASS VALLEY CALIFORNIA

## Goals:

- To oxidize and precipitate the iron and co precipitate arsenic from solution.
- It was initially proposed that sodium hydroxide would be added to raise the pH from 6.6 approximately 8.0.
- The addition of base was not necessary (Degassing of carbon dioxide from the water)



# Treatment Schematic RCTS 4 Rotor Empire Mine





# **EMPIRE MINE GRASS VALLEY CALIFORNIA**

## **Results:**

**Iron concentrations were reduced from 4290  $\mu\text{g/L}$  to 80  $\mu\text{g/L}$ .  
(without base addition)**

**The ferrous iron concentration was not sufficient to co-precipitate all of the arsenic from solution. Arsenic was reduced from 47  $\mu\text{g/L}$  to 25  $\mu\text{g/L}$ .**

**Suggested adding Ferrous iron to co-precipitate arsenic**

# Summary

- **Can be rapidly mobilized.**
- **Efficient lime utilization**
- **Can reduce sludge production.**
- **Requires 60% to 90% less expended energy than conventional treatment.**
- **Less space required.**
- **Can treat as a batch or continuous.**