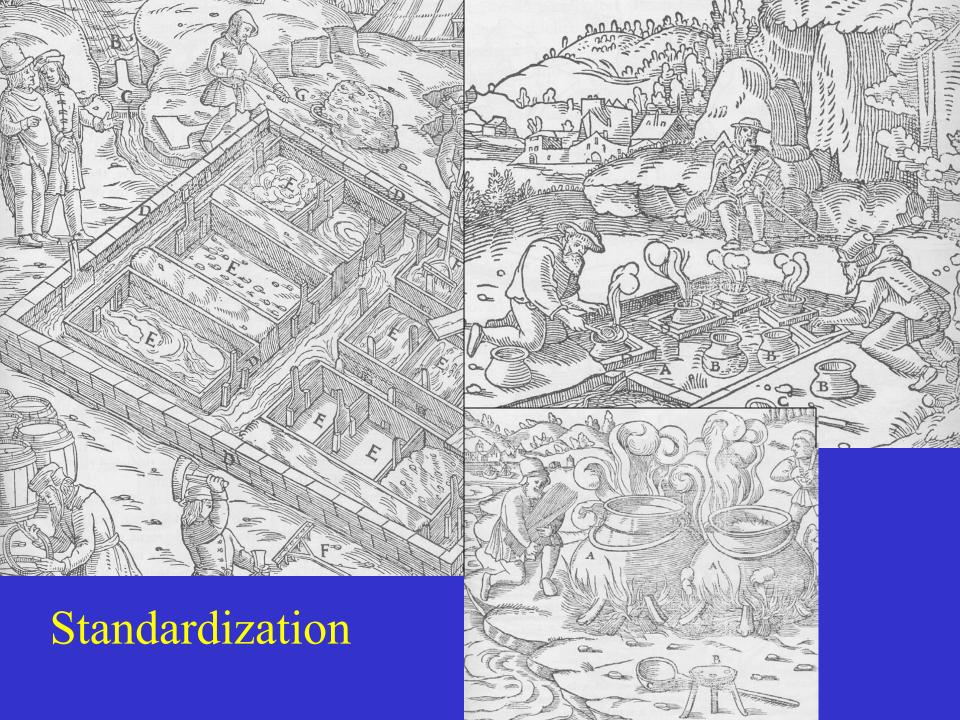


## **Objectives**

- 1. Standardize procedures
- 2. Improve method
  - a. CO<sub>2</sub>-enriched air
  - b. Quantify particle size effects
- 3. Flexibility in method
- 4. Improved interpretations

Sought a Variety of Rocks to Leach





#### GEORGIUS AGRICOLA

## DE RE METALLICA

#### TRANSLATED FROM THE FIRST LATIN EDITION OF 1556

with

Biographical Introduction, Annotations and Appendices upon the Development of Mining Methods, Metallurgical Processes, Geology, Mineralogy & Mining Law from the earliest times to the 16th Century

BY

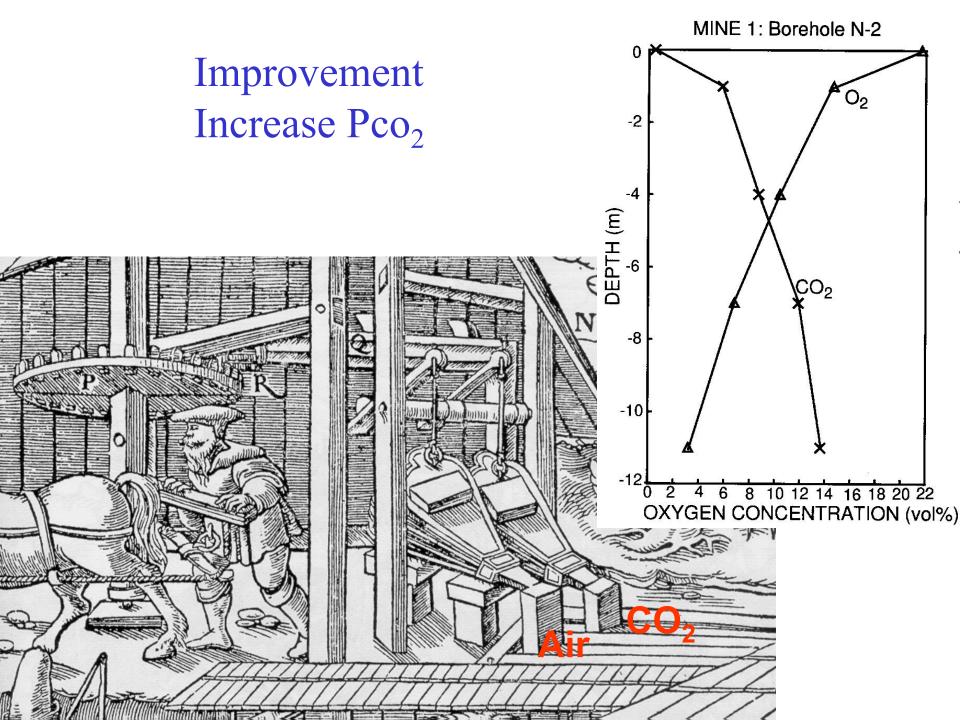
#### HERBERT CLARK HOOVER

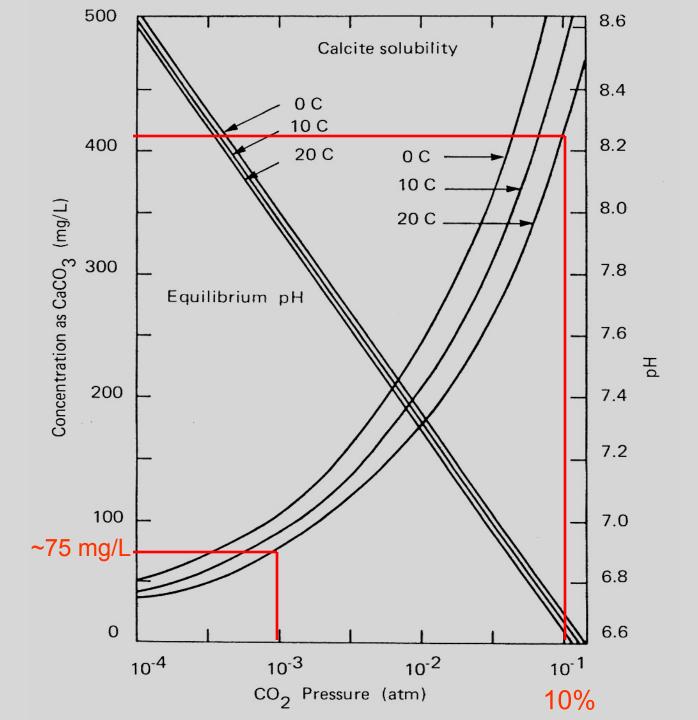
A. B. Stanford University, Member American Institute of Mining Engineers, Mining and Metallurgical Society of America, Société des Ingénieurs Civils de France, American Institute of Civil Engineers, Fellow Royal Geographical Society, etc., etc.

AND

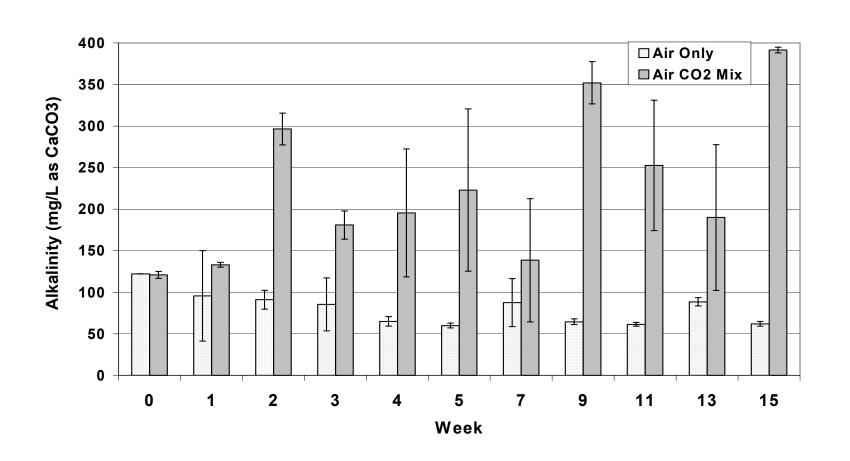
#### LOU HENRY HOOVER

A. B. Stanford University, Member American Association for the Advancement of Science, The National Geographical Society, Royal Scottish Geographical Society, etc., etc.

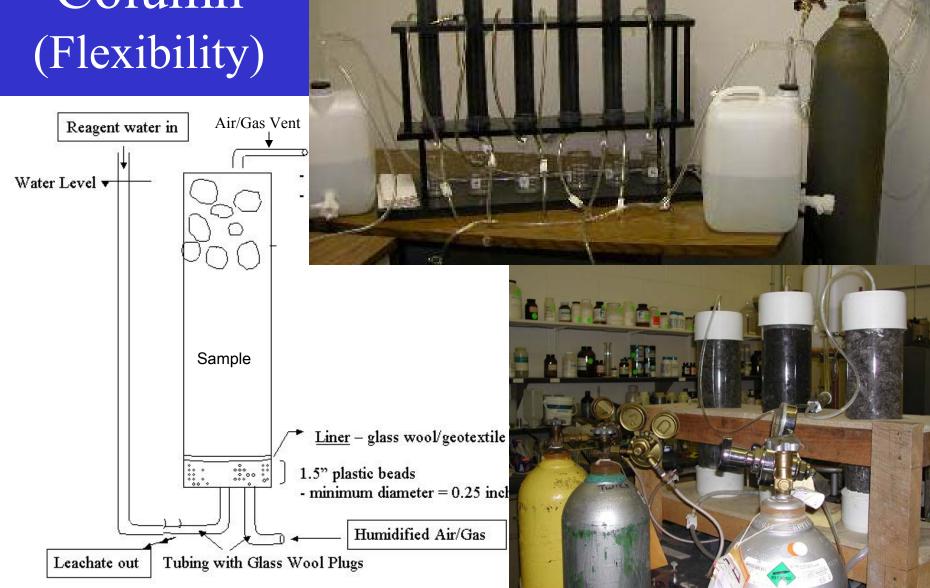


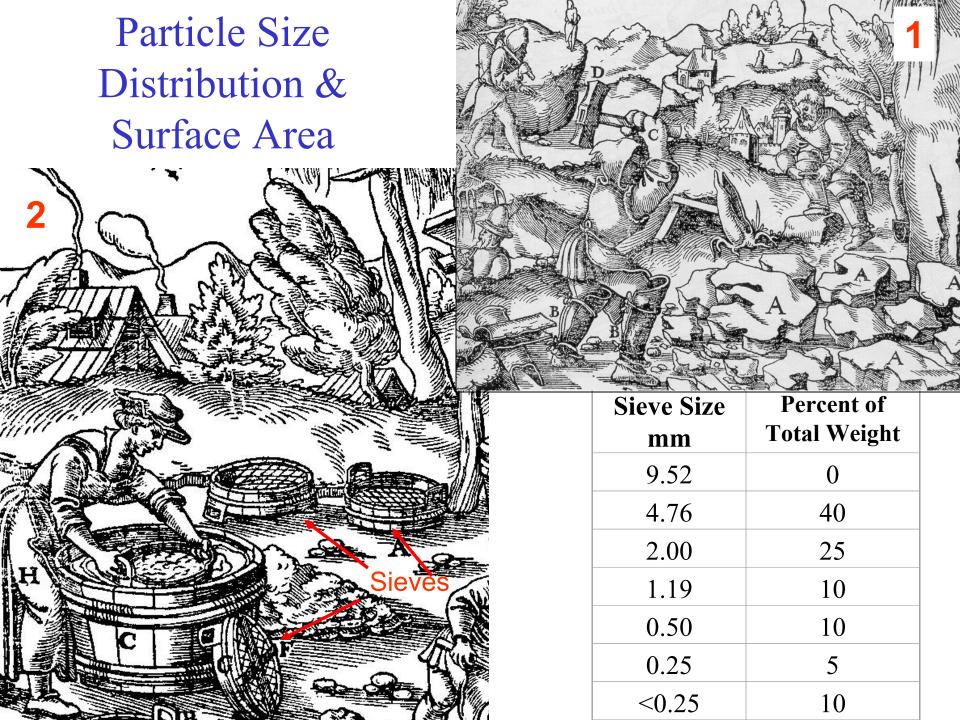


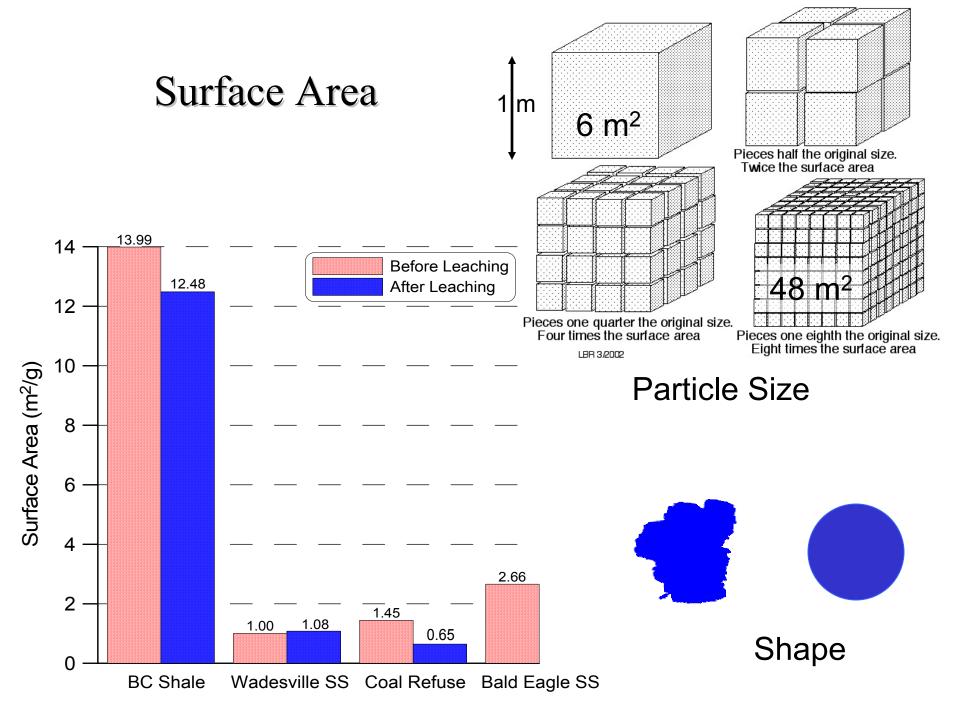
## Results from 2002: CO<sub>2</sub> Mix, Higher Alkalinity

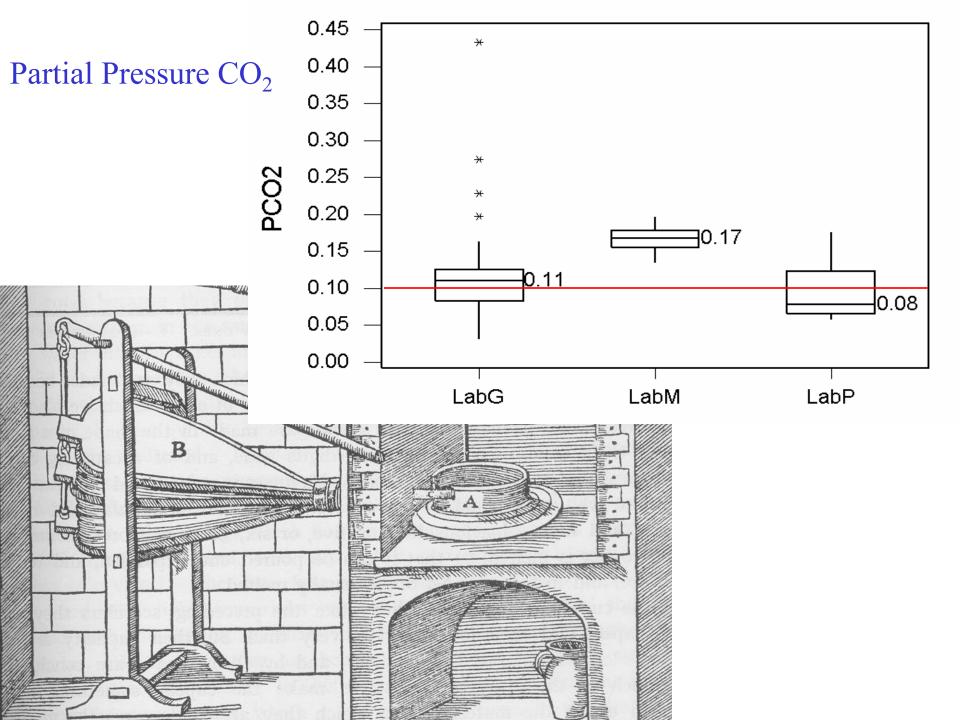


# Leaching Column (Flexibility)





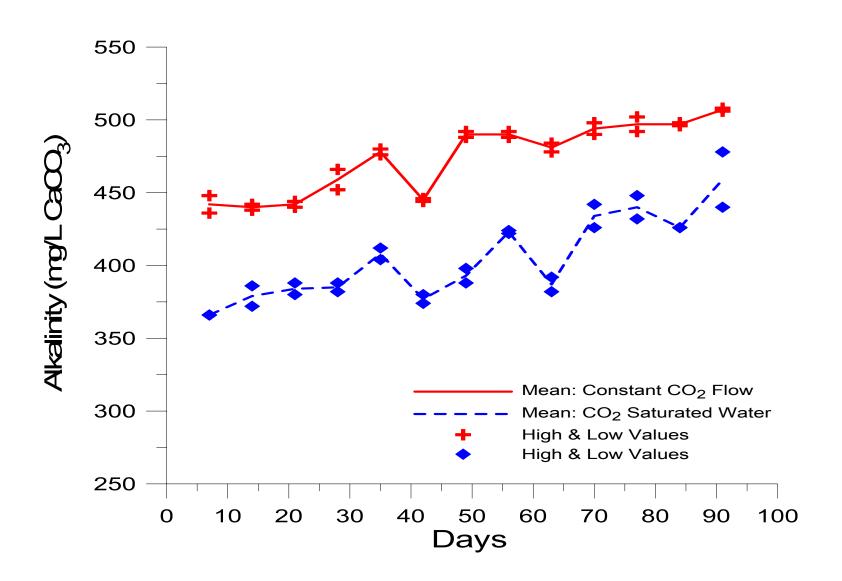


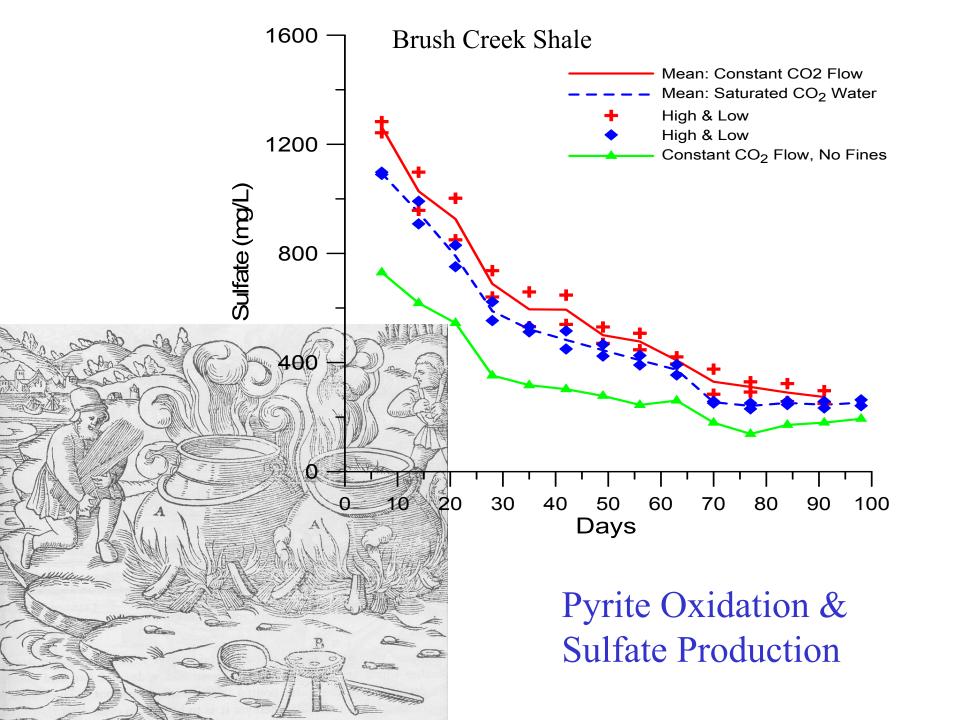


## **Interpretations**



## Carbonate Dissolution & Alkalinity Generation

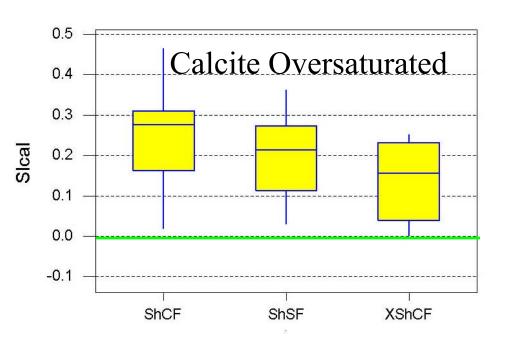




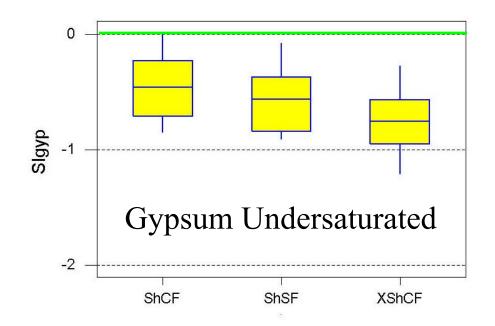
# Cumulative Load (Mass) of CaCO<sub>3</sub> & Sulfur (or How much weathered each week?)

- 1. Calculated Mass of Sulfur and CaCO<sub>3</sub> in Column using total percent sulfur (%S) and neutralization potential (NP)
- 2. Determined amount of **sulfur** weathered from: sulfate x volume of leachate\*
- 3. Determined amount of CaCO<sub>3</sub> weathered from: (HCO<sub>3</sub> + alkalinity neutralized) x volume of leachate.
  - Alkalinity neutralized calculated from SO<sub>4</sub>\*

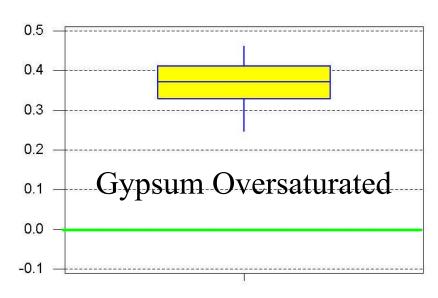
<sup>\*</sup> For this to work must be undersaturated w/ respect to gypsum



**Brush Creek Shale** 

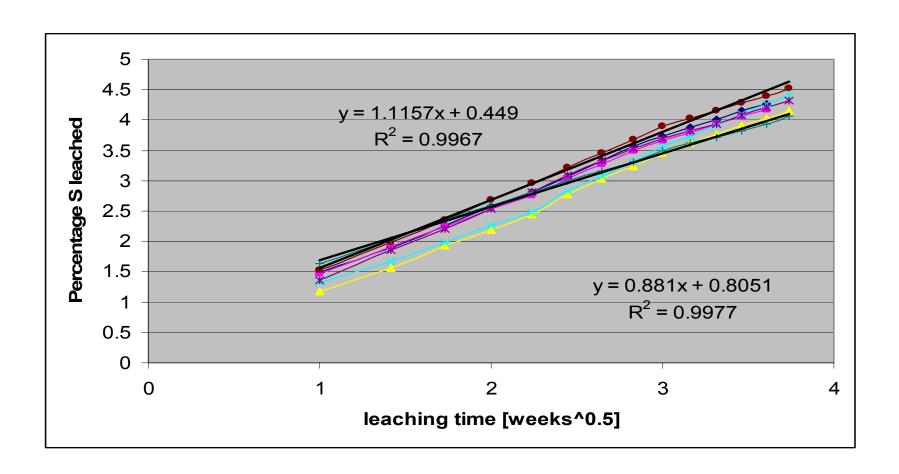


## Effects of Solubility on Interpretations

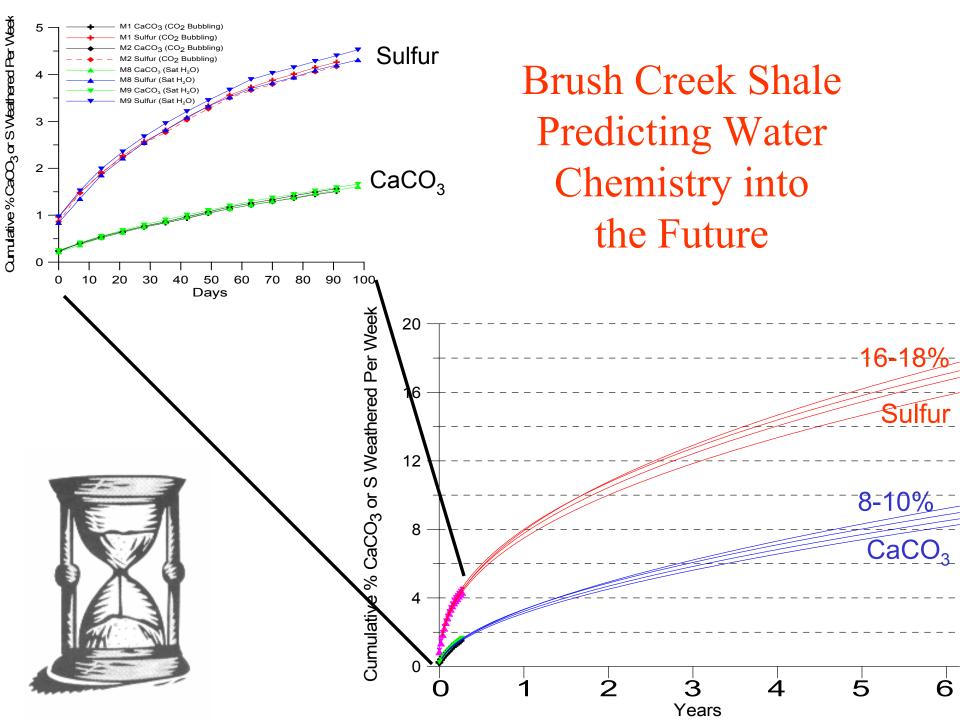


**Determine Saturation Indices** 

### Brush Creek Pyrite Oxidation Diffusion Reaction



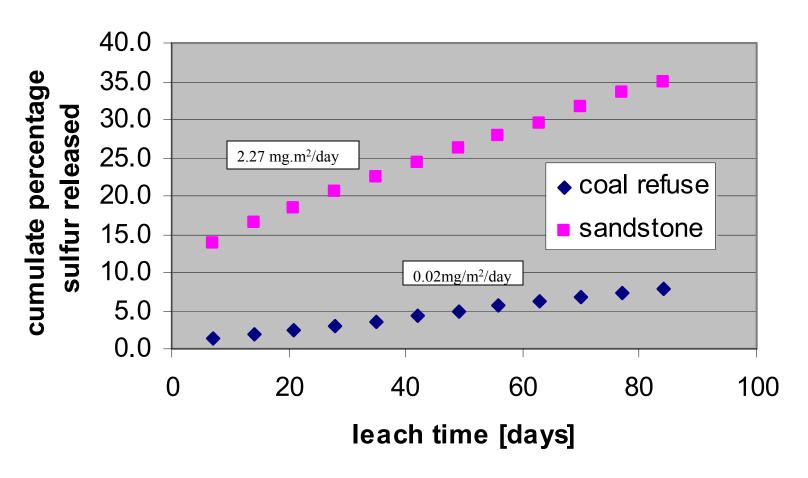
$$N_{s} = Kt^{1/2} + F$$







#### Weathering Rates of Coal Refuse & Hydrothermal Pyrite



## **Findings**

- 1. The ADTI-WP2 method produces concentrations of acidity, alkalinity, sulfates and metals comparable to mine environment.
- 2. The laboratory performance show good repeatability for duplicate samples in columns.
- 3. It is important to normalize the leaching data. Steps include: standardized particle size distribution; surface areas measurements, and surface area to volume ratios.

## Findings, continued

- 4. %S leached through time for the different rocks indicate differences in leaching processes. The Brush Creek shale exhibits a square root of time dependence, which suggests a diffusion-controlled process. The coal refuse and the Bald Eagle sandstone conform to a simple linear weathering process.
  - 5. To interpret leaching data solubility controls must be considered.

