

In-Situ Bioremediation Demonstration of Coal-Based Acid Mine Drainage

Tide Mine Site Indiana County, Pennsylvania

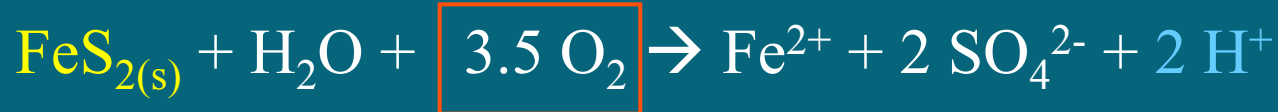
Presented to:
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Symposium

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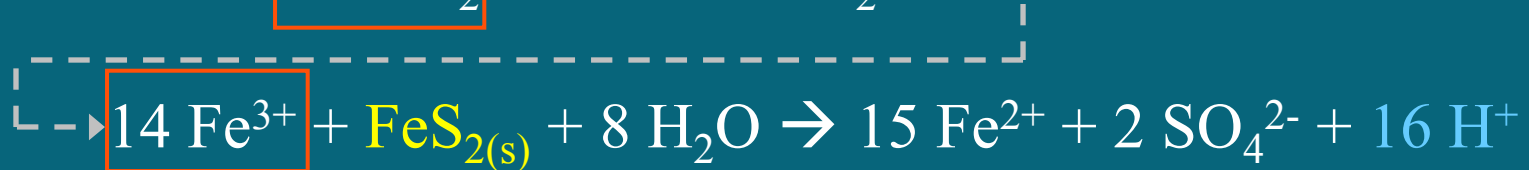
20 April 2005

AMD Chemistry

Initiation Reaction:



Propagation Reactions:

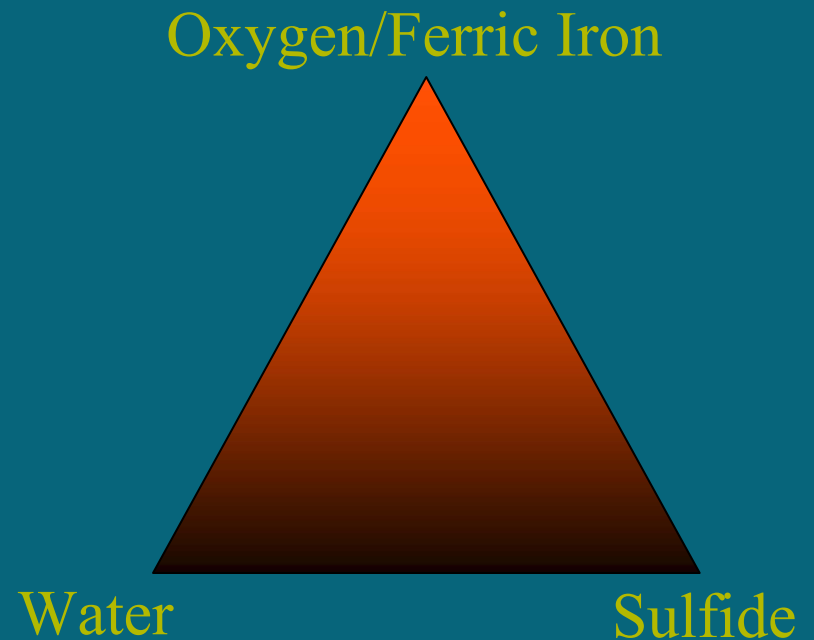


(Singer & Stumm, 1970)

In-situ treatment can eliminate the causative agents of AMD

Breaking the AMD Triangle

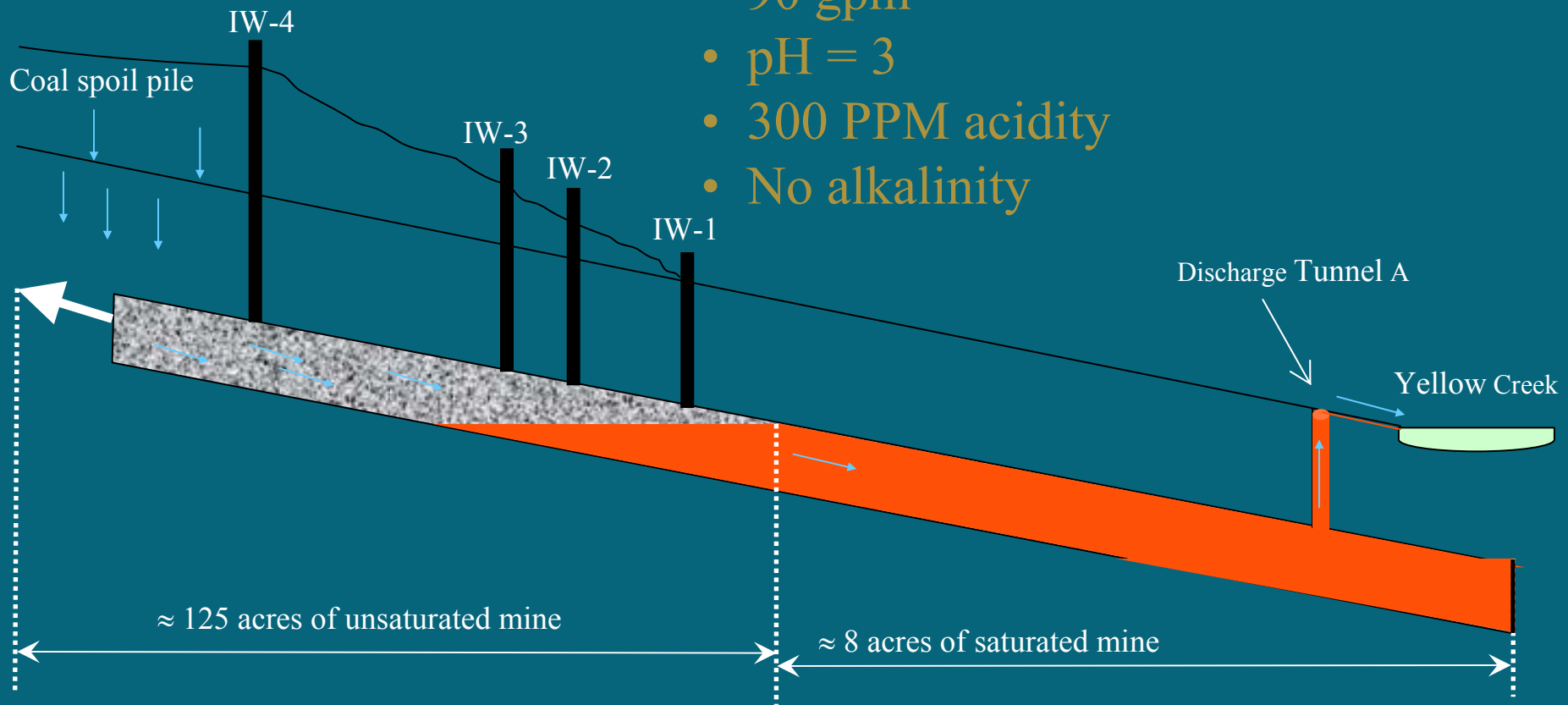
- Provide an alternative sink for oxygen and ferric iron (e.g., a carbon source)
- Oxidation of the carbon source generates carbon dioxide/alkalinity
- Eliminating/minimizing water flushing - not required



Tide Mine Schematic

Flow to Yellow Creek:

- 90 gpm
- pH = 3
- 300 PPM acidity
- No alkalinity

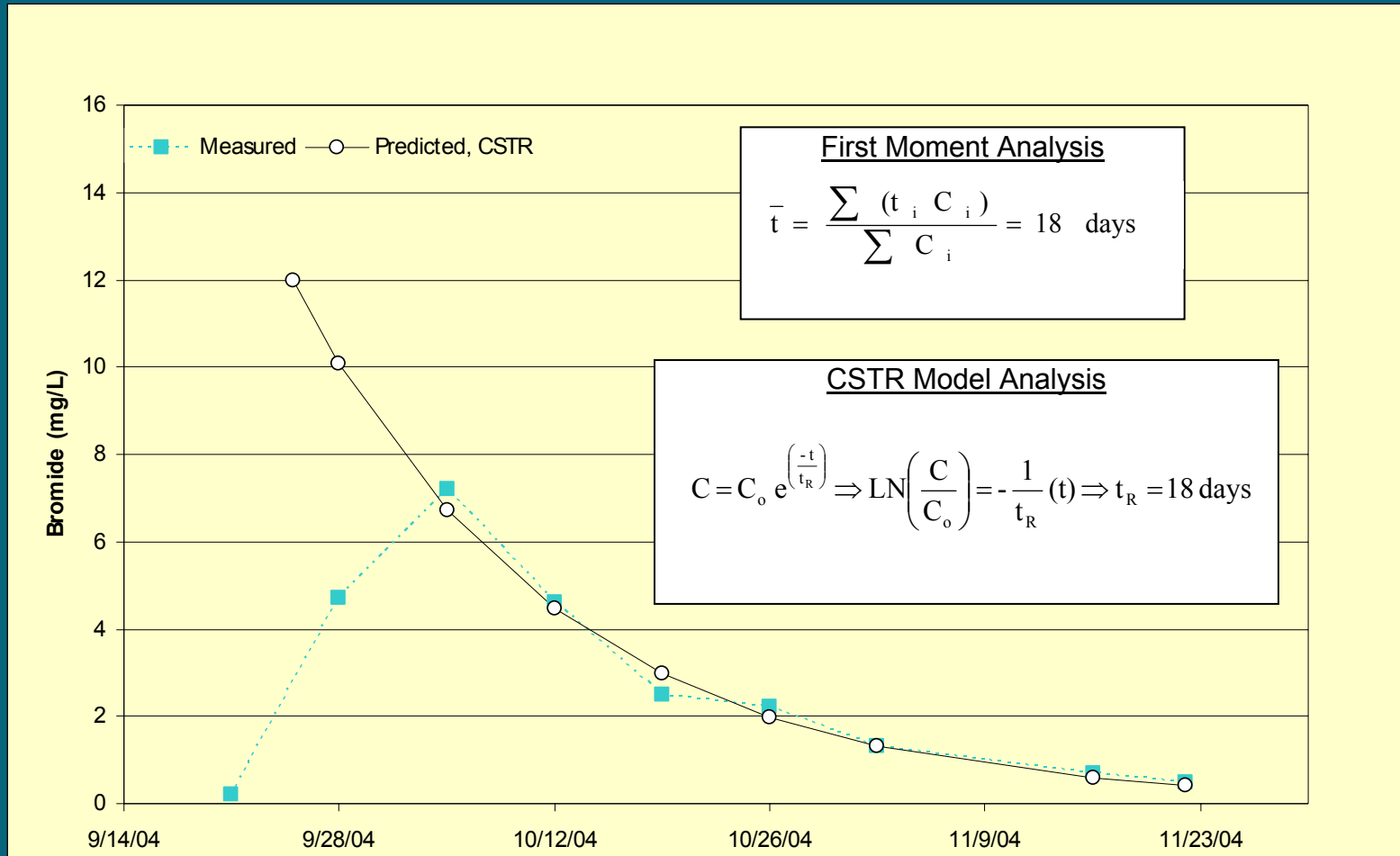


Demonstration Project Plan

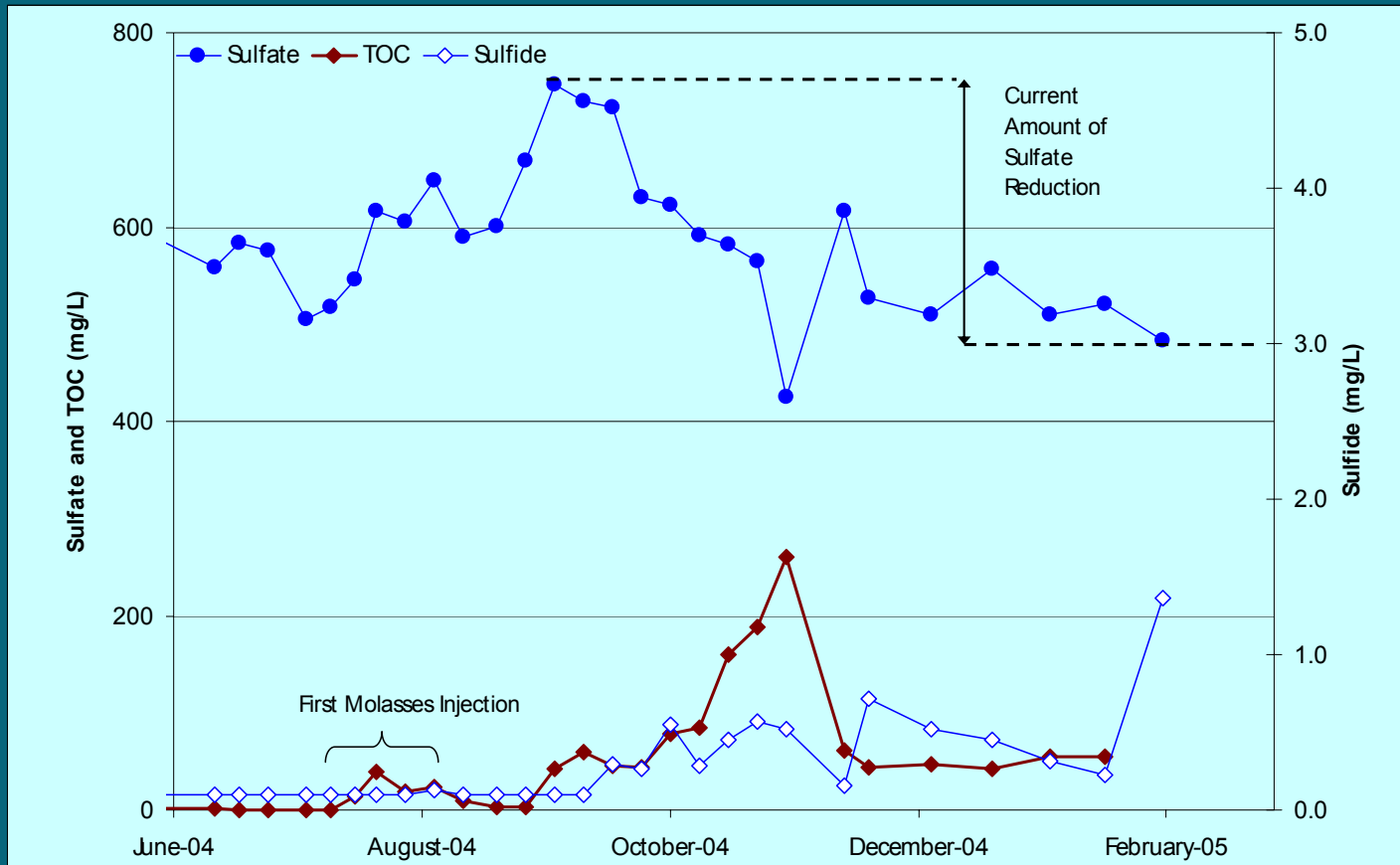
- 3 months of recirculation system operation with caustic and molasses injection to affect net change in biogeochemistry
- Bromide tracer study to characterize mine pool hydraulics
- Carbon dioxide gas injection to alter mine atmosphere
- Additional 3 months of passive methanol injection to demonstrate sustainability of in-situ treatment



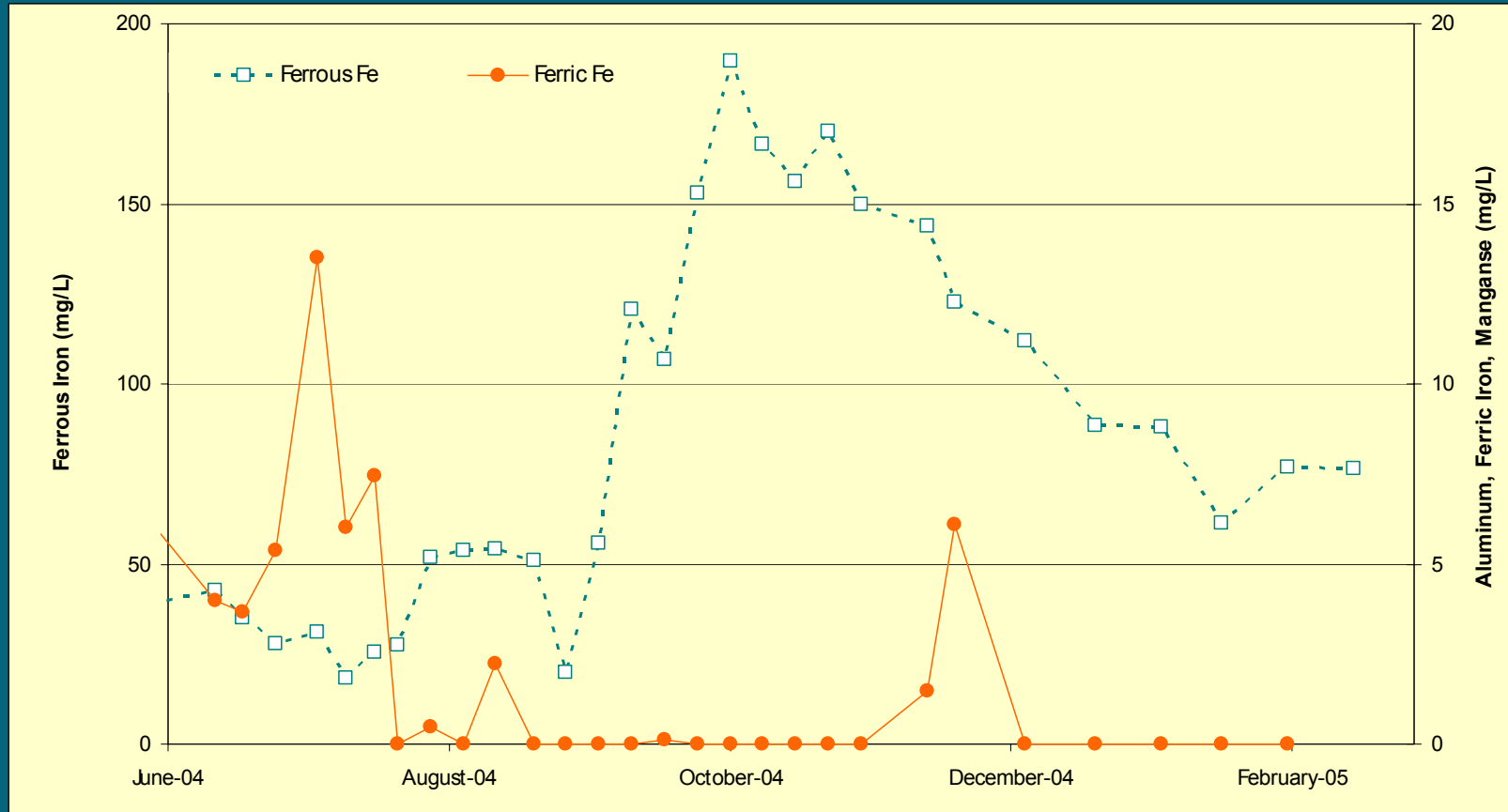
Bromide Tracer Test to Evaluate Mine Pool Hydraulics



Elevated TOC Results in Sulfate Reducing Conditions



Metals Performance Monitoring Data

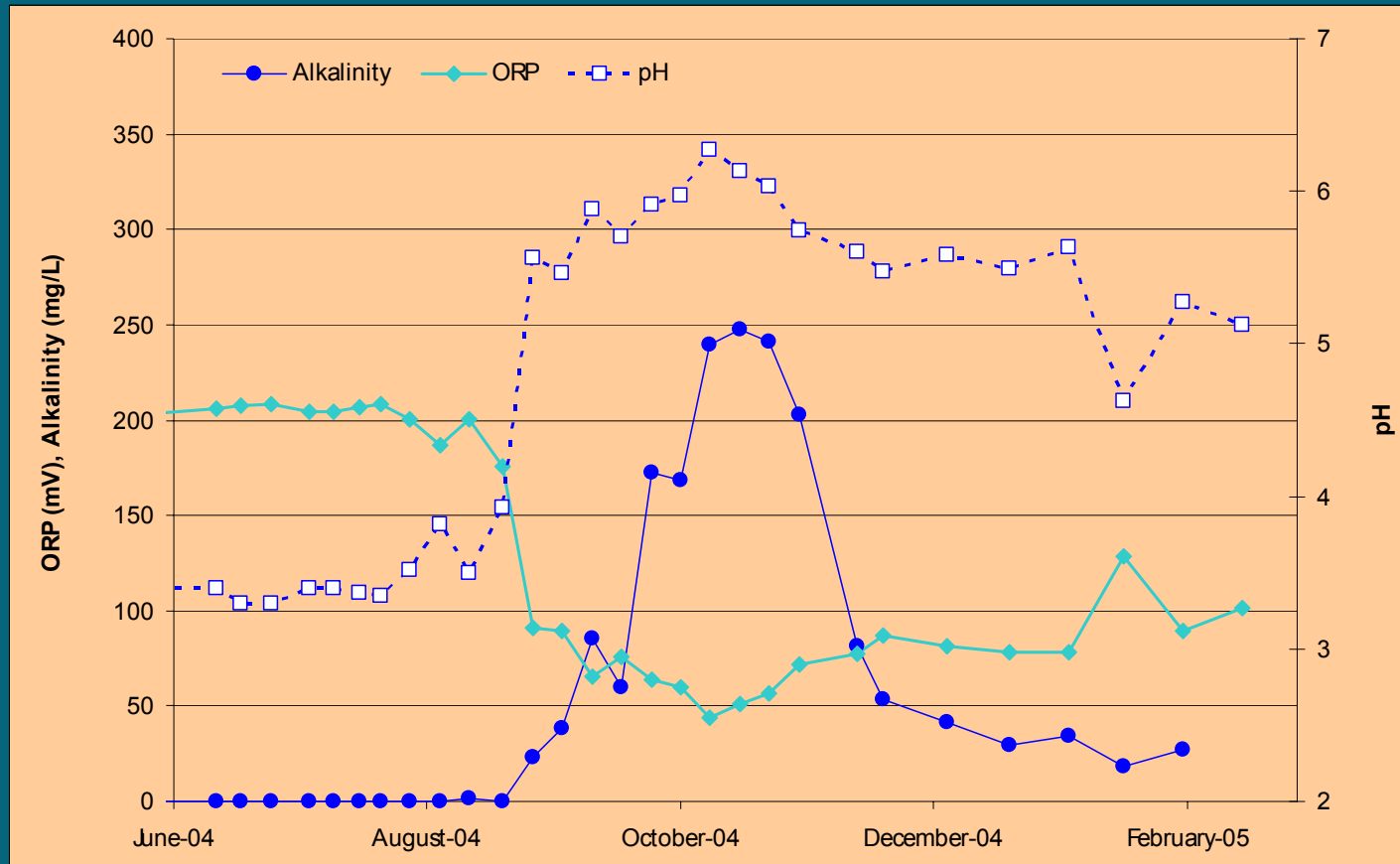


Aluminum concentrations decreased greater than 50% during treatment – function of pH

Continuing Sources of Iron in Mine Workings

- Reductive dissolution of solid-phase ferric hydroxide in the saturated portion of the mine workings
 - Controlled through FeS precipitation and maintaining sulfate reduction in the mine pool
 - Saturated portion accounts for approximately 6% of mine workings
- AMD-impacted water from the unsaturated portions of the mine workings
 - Gas phase oxygen provides continuous source of AMD causative agents (i.e., dissolved oxygen and ferric iron)
 - Requires gas phase treatment for AMD prevention or continuous liquid phase treatment for control
 - Unsaturated portion accounts for approximately 94% of mine workings

Carbonate Alkalinity Is Now Dominant In Previously Acidic Mine Pool



Re-Oxidation of Observed FeS?

In-Situ Treatment: $\text{FeS} + 2\text{O}_2 \rightarrow \text{FeSO}_4 + \text{no acidity}$

Natural Aerobic State: $\text{FeS}_2 + 3.5\text{O}_2 + \text{H}_2\text{O} \rightarrow \text{FeSO}_4 + \text{H}_2\text{SO}_4$

At the same operating re-dox state, oxidation of ferrous sulfides will not generate acidity in contrast with pyritic sulfides. Consistent with this expectation, alkalinity is observed from Tide Mine during the treatment period after washout of the NaOH neutralization effect

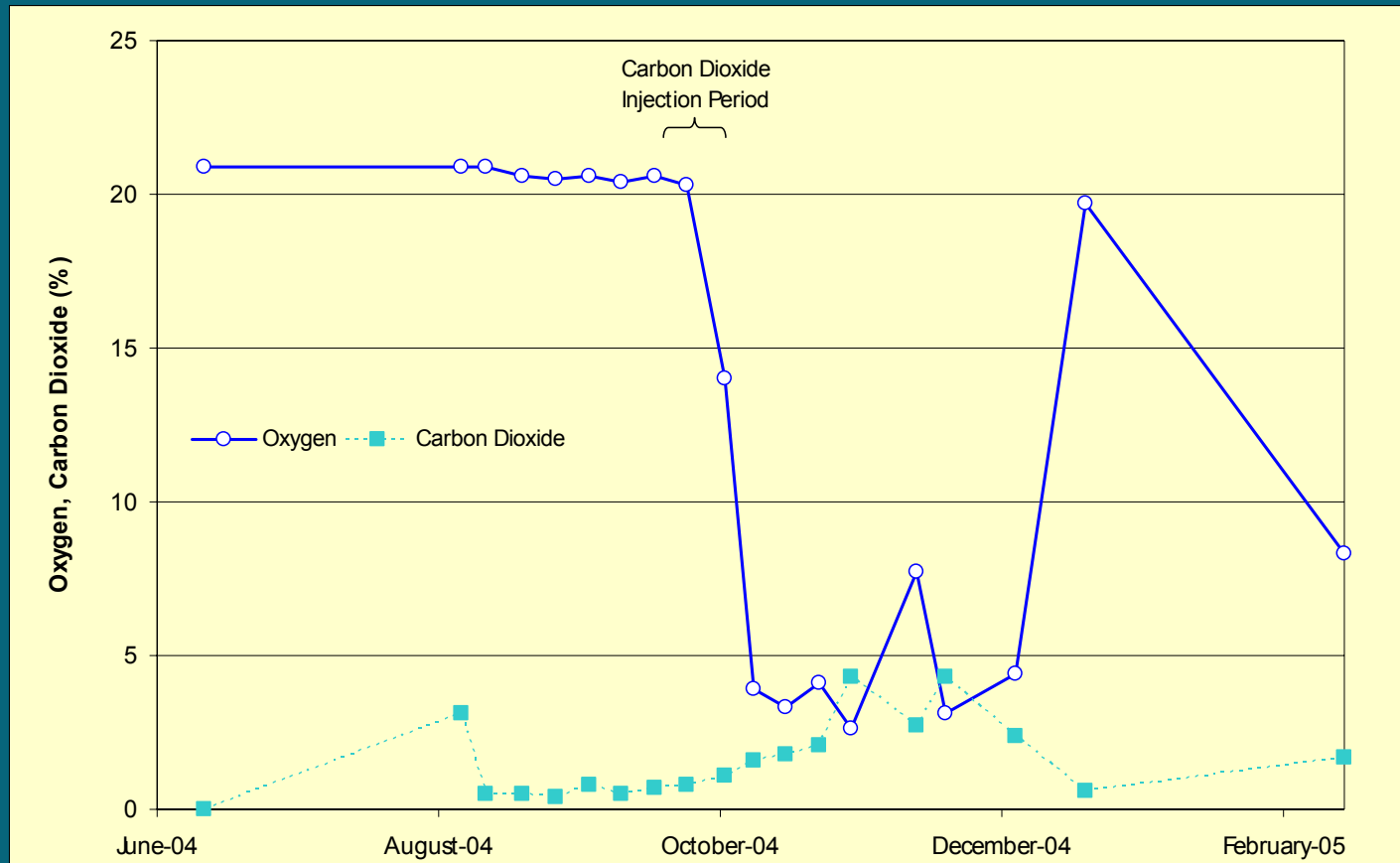


Carbon Dioxide Injection - Why Control the Gas Phase?

Oxygen Molar Balance

- Pyrite oxidation stoichiometry indicates $O_2:SO_4$ is 2:1
- Oxygen-saturated water holds 8 ppm or 0.25 mM O_2
- Assuming full conversion, 0.25 mM of O_2 creates 0.125 mM SO_4
- Tide mine water has ≈ 8 mM sulfate
- Therefore, at a maximum, 1.5% of the sulfate produced by the mine is generated by aqueous phase oxygen
- 98.5% of the sulfate (and acidity) is being generated by gas phase molecular oxygen
- With O_2 consumed by AMD reactions, resulting gas is buoyant – maintaining 5% CO_2 through direct injection or biological activity generates a dense anaerobic blanket

CO₂ addition to the mine atmosphere broke the Oxygen “supply lines”



Results

- Tide Mine project demonstrated the efficacy of in-situ treatment – proof of concept
- No dissolved oxygen in the mine pool or ferric iron in the mine pool
- Sustained increased pH in the mine pool, continued alkalinity detection
- Decreasing overall iron concentrations, reduced aluminum concentration
- Gas exchange occurs over large footprint of unsaturated mine workings; therefore, sustained CO₂ injection required for gas phase control at Tide Mine

How Does In-Situ Treatment Fit with the “Big Picture”?

- Economics of prevention vs. collection and treatment
- System-wide application of in-situ technology to stop or reduce AMD, minimize risk with blowout or mine failure
- Low cost O&M system implemented in-situ
 - Sludge settling occurs in mine workings
 - Use of local agricultural products and waste streams
- Compatible with passive polishing systems